IMMEDIATELY DANGEROUS to LIFE or HEALTH

Bromine Trifluoride CAS[®] No. 7787-71-5

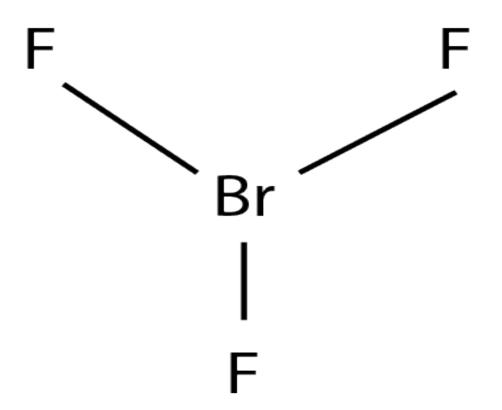


Centers for Disease Control and Prevention National Institute for Occupational Safety and Health This page intentionally left blank.

IMMEDIATELY DANGEROUS TO LIFE OR HEALTH (IDLH) VALUE PROFILE

BROMINE TRIFLUORIDE

[CAS® NO. 7787-71-5]



DEPARTMENT OF HEALTH AND HUMAN SERVICES Centers for Disease Control and Prevention National Institute for Occupational Safety and Health This document is in the public domain and may be freely copied or reprinted.

Disclaimer

Mention of any company or product does not constitute endorsement by the National Institute for Occupational Safety and Health, Centers for Disease Control and Prevention. In addition, citations to websites external to NIOSH do not constitute NIOSH endorsement of the sponsoring organizations or their programs or products. Furthermore, NIOSH is not responsible for the content of these websites. All web addresses referenced in this document were accessible as of the publication date.

Get More Information

Find NIOSH products and get answers to workplace safety and health questions:

1-800-CDC-INFO (1-800-232-4636) | TTY: 1-888-232-6348 CDC/NIOSH INFO: cdc.gov/info | cdc.gov/niosh Monthly *NIOSH eNews*: cdc.gov/niosh/eNews

Suggested Citation

NIOSH [2020]. Immediately dangerous to life or health (IDLH) value profile: bromine trifluoride, CAS[®] No. 7787-71-5. By Niemeier RT. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2020-123.

DOI: https://doi.org/10.26616/NIOSHPUB2020123

DHHS (NIOSH) Publication No. 2020-123

July 2020

Foreword

Chemicals are a ubiquitous component of the modern workplace. Occupational exposures to chemicals have the potential to adversely affect the health and lives of workers. Acute or short-term exposures to high concentrations of some airborne chemicals can quickly overwhelm workers, resulting in a spectrum of undesirable health outcomes that may inhibit the ability to escape from the exposure environment (e.g., irritation of the eyes and respiratory tract or cognitive impairment), cause severe irreversible effects (e.g., damage to the respiratory tract or reproductive toxicity), and in extreme cases, cause death. Airborne concentrations of chemicals capable of causing such adverse health effects or of impeding escape from high-risk conditions may arise from a variety of non-routine workplace situations, including special work procedures (e.g., in confined spaces), industrial accidents (e.g., chemical spills or explosions), and chemical releases into the community (e.g., during transportation incidents or other uncontrolled-release scenarios).

The immediately dangerous to life or health (IDLH) air concentration values developed by the National Institute for Occupational Safety and Health (NIOSH) characterize these highrisk exposure concentrations and conditions [NIOSH 2013]. IDLH values are based on a 30-minute exposure duration and have traditionally served as a key component of the decision logic for the selection of respiratory protection devices [NIOSH 2004].

Occupational health professionals have employed these values beyond their initial purpose as a component of the *NIOSH Respirator Selection Logic* to assist in developing risk management plans for non-routine work practices governing operations in high-risk environments (e.g., confined spaces) and the development of emergency preparedness plans.

The approach used to derive IDLH values for high priority chemicals is outlined in the *NIOSH Current Intelligence Bulletin (CIB)* 66: *Derivation of Immediately Dangerous to Life or Health (IDLH) Values* [NIOSH 2013]. CIB 66 provides (1) an update on the scientific basis and risk assessment methodology used to derive IDLH values, (2) the rationale and derivation process for IDLH values, and (3) a demonstration of the derivation of scientifically credible IDLH values using available data resources.

This technical report presents the IDLH value for bromine trifluoride (CAS[®] No. 7787-71-5). The scientific basis, toxicologic data, and risk assessment approach used to derive the IDLH value are summarized to ensure transparency and scientific credibility.

John Howard, M.D. Director National Institute for Occupational Safety and Health Centers for Disease Control and Prevention This page intentionally left blank.

Contents

Foreword	iii
Abbreviations	vi
Glossary	viii
Acknowledgments	xiii
1 Introduction	1
1.1 Overview of the IDLH Value for Bromine Trifluoride	1
1.2 Purpose	1
1.3 Literature Search Strategy and Data Evaluation	1
1.4 General Substance Information	1
2 Human Data	4
3 Animal Toxicity Data	5
4 Summary	9
5 References	10
Appendix A—Literature Search Strategy and Data Evaluation	12

Abbreviations

ACGIH [®]	American Conference of Governmental Industrial Hygienists
Acuate RfC	Acute reference concentration
AEGLs	Acute Exposure Guideline Levels
AIHA®	American Industrial Hygiene Association
BMC	benchmark concentration
BMD	benchmark dose
BMR	benchmark response
BMCL	benchmark concentration lower confidence limit
BrF ₃	Bromine trifluoride
BrF ₅	Bromine pentafluoride
°C	degrees Celsius
CAS®	Chemical Abstracts Service, a division of the American Chemical Society
CIB	Current Intelligence Bulletin
ClF ₃	Chlorine trifluoride
ClF ₅	Chlorine pentafluoride
EPA	Environmental Protection Agency
ERPGs™	Emergency Response Planning Guidelines
ET ₅₀	Effective time to 50% mortality
°F	degrees Fahrenheit
g/cu cm	grams per cubic centimeter
HF	Hydrogen fluoride
IDLH	immediately dangerous to life or health
LC	lethal concentration
LC ₅₀	median lethal concentration
LC_{01}	lethal concentration 1%
LC	lowest concentration that caused death in humans or animals
LD_{50}	lethal dose
LD _{LO}	lowest dose that caused death in humans or animals
LEL	lower explosive limit
LOAEL	lowest observed adverse effect level
mg/m ³	milligram(s) per cubic meter
min	minutes
mmHg	millimeter(s) of mercury
MSHA	Mine Safety and Health Administration
NAS	National Academy of Sciences
NIOSH	National Institute for Occupational Safety and Health
NLM	National Library of Medicine
NOAEL	no observed adverse effect level
NRC	National Research Council

OEL	occupational exposure limit
OSHA	Occupational Safety and Health Administration
PEL	permissible exposure limit
POD	point of departure
ppm	parts per million
RD ₅₀	concentration of a chemical in the air that is estimated to cause a 50%
	decrease in the respiratory rate
REL	recommended exposure limit
STEL	short-term exposure limit
TERA	Toxicology Excellence for Risk Assessment
TLV®	Threshold Limit Value
TWA	time-weighted average
UEL	upper explosive limit
UF	uncertainty factor
WEELs®	Workplace Environmental Exposure Levels

Glossary

Acute exposure: Exposure by the oral, dermal, or inhalation route for 24 hours or less.

Acute Exposure Guideline Levels (AEGLs): Threshold exposure limits for the general public, applicable to emergency exposure periods ranging from 10 minutes to 8 hours. AEGL-1, AEGL-2, and AEGL-3 are developed for five exposure periods (10 and 30 minutes, 1 hour, 4 hours, and 8 hours) and are distinguished by varying degrees of severity of toxic effects, ranging from transient, reversible effects to life-threatening effects [NRC 2001]. AEGLs are intended to be guideline levels used during rare events or single once-in-a-lifetime exposure to airborne concentrations of acutely toxic, high-priority chemicals [NRC 2001]. The threshold exposure limits are designed to protect the general population, including the elderly, children, and other potentially sensitive groups that are generally not considered in the development of workplace exposure recommendations (additional information available at https://www.epa.gov/oppt/aegl/).

Acute reference concentration (Acute RfC): An estimate (with uncertainty spanning perhaps an order of magnitude) of a continuous inhalation exposure for an acute duration (24 hours or less) of the human population (including sensitive subgroups) that is likely to be without an appreciable risk of deleterious effects during a lifetime. It can be derived from a NOAEL, LOAEL, or benchmark concentration, with uncertainty factors (UFs) generally applied to reflect limitations of the data used. Generally used in EPA noncancer health assessments [EPA 2019].

Acute toxicity: Any poisonous effect produced within a short period of time following an exposure, usually 24 to 96 hours [EPA 2019].

Adverse effect: A substance-related biochemical change, functional impairment, or pathologic lesion that affects the performance of an organ or system or alters the ability to respond to additional environmental challenges.

Benchmark dose/concentration (BMD/BMC): A dose or concentration that produces a predetermined change in response rate of an effect (called the benchmark response, or BMR) compared with background [EPA 2019] (additional information available at https://www.epa. gov/ncea/bmds/).

Benchmark response (BMR): An adverse effect, used to define a benchmark dose from which a reference dose or concentration can be developed. The change in response rate over background of the BMR is usually in the range of 5–10%, which is the limit of responses typically observed in well-conducted animal experiments [EPA 2019].

BMCL: A statistical lower confidence limit on the concentration at the BMC [EPA 2019].

Bolus exposure: A single, relatively large dose.

Ceiling value (**"C"**): Term in occupational exposure indicating the airborne concentration of a potentially toxic substance that should never be exceeded in a worker's breathing zone.

Chronic exposure: Repeated exposure for an extended period of time. Typically, exposures are more than approximately 10% of life span for humans and >90 days to 2 years for laboratory species.

Critical study: The study that contributes most significantly to the qualitative and quantitative assessment of risk [EPA 2019].

Dose: The amount of a substance available for interactions with metabolic processes or biologically significant receptors after crossing the outer boundary of an organism [EPA 2019].

 ECt_{50} : A combination of the effective concentration of a substance in the air and the exposure duration that is predicted to cause an effect in 50% of the experimental test subjects.

Emergency Response Planning Guidelines (ERPGsTM): Maximum airborne concentrations below which nearly all individuals can be exposed without experiencing health effects for 1-hour exposure. ERPGs are presented in a tiered fashion, with health effects ranging from mild or transient to serious, irreversible, or life threatening (depending on the tier). ERPGs are developed by the American Industrial Hygiene Association [AIHA 2016].

Endpoint: An observable or measurable biological event or sign of toxicity, ranging from biomarkers of initial response to gross manifestations of clinical toxicity.

Exposure: Contact made between a chemical, physical, or biological agent and the outer boundary of an organism. Exposure is quantified as the amount of an agent available at the exchange boundaries of the organism (e.g., skin, lungs, gut).

Extrapolation: An estimate of the response at a point outside the range of the experimental data, generally through the use of a mathematical model, although qualitative extrapolation may also be conducted. The model may then be used to extrapolate to response levels that cannot be directly observed.

Hazard: A potential source of harm. Hazard is distinguished from risk, which is the probability of harm under specific exposure conditions.

Immediately dangerous to life or health (IDLH) condition: A condition that poses a threat of exposure to airborne contaminants when that exposure is likely to cause death or immediate or delayed permanent adverse health effects or prevent escape from such an environment [NIOSH 2004, 2013].

IDLH value: A maximum (airborne concentration) level above which only a highly reliable breathing apparatus providing maximum worker protection is permitted [NIOSH 2004, 2013]. IDLH values are based on a 30-minute exposure duration.

 LC_{01} : The statistically determined concentration of a substance in the air that is estimated to cause death in 1% of the test animals.

 LC_{50} : The statistically determined concentration of a substance in the air that is estimated to cause death in 50% of the test animals; median lethal concentration.

 LC_{L0} : The lowest lethal concentration of a substance in the air reported to cause death, usually for a small percentage of the test animals.

 LD_{50} : The statistically determined lethal dose of a substance that is estimated to cause death in 50% of the test animals; median lethal concentration.

 LD_{LO} : The lowest dose of a substance that causes death, usually for a small percentage of the test animals.

Lethality: Pertaining to or causing death; fatal; referring to the deaths resulting from acute toxicity studies. May also be used in lethality threshold to describe the point of sufficient substance concentration to begin to cause death.

Lower explosive limit (LEL): The minimum concentration of a gas or vapor in air, below which propagation of a flame does not occur in the presence of an ignition source.

Lowest observed adverse effect level (LOAEL): The lowest tested dose or concentration of a substance that has been reported to cause harmful (adverse) health effects in people or animals.

Mode of action: The sequence of significant events and processes that describes how a substance causes a toxic outcome. By contrast, the term "*mechanism of action*" implies a more detailed understanding on a molecular level.

No observed adverse effect level (NOAEL): The highest tested dose or concentration of a substance that has been reported to cause no harmful (adverse) health effects in people or animals.

Occupational exposure limit (OEL): Workplace exposure recommendations developed by governmental agencies and nongovernmental organizations. OELs are intended to represent the maximum airborne concentrations of a chemical substance below which workplace exposures should not cause adverse health effects. OELs may apply to ceiling limits, STELs, or TWA limits.

Peak concentration: Highest concentration of a substance recorded during a certain period of observation.

Permissible exposure limits (PELs): Occupational exposure limits developed by OSHA (29 CFR 1910.1000) or MSHA (30 CFR 57.5001) for allowable occupational airborne exposure concentrations. PELs are legally enforceable and may be designated as ceiling limits, STELs, or TWA limits.

Point of departure (POD): The point on the dose–response curve from which dose extrapolation is initiated. This point can be the lower bound on dose for an estimated incidence or a change in response level from a concentration-response model (BMC), or it can be a NOAEL or LOAEL for an observed effect selected from a dose evaluated in a health effects or toxicology study.

 RD_{50} : The statistically determined concentration of a substance in the air that is estimated to cause a 50% decrease in the respiratory rate.

Recommended exposure limit (REL): Recommended maximum exposure limit to prevent adverse health effects, based on human and animal studies and established for occupational (up to 10-hour shift, 40-hour week) inhalation exposure by NIOSH. RELs may be designated as ceiling limits, STELs, or TWA limits.

Short-term exposure limit (STEL): A worker's 15-minute time-weighted average exposure concentration that shall not be exceeded at any time during a work day.

Target organ: Organ in which the toxic injury manifests in terms of dysfunction or overt disease.

Threshold Limit Values (**TLVs**[®]): Recommended guidelines for occupational exposure to airborne contaminants, published by the American Conference of Governmental Industrial Hygienists (ACGIH[®]). TLVs[®] refer to airborne concentrations of chemical substances and represent conditions under which it is believed that nearly all workers may be repeatedly exposed, day after day, over a working lifetime, without adverse effects. TLVs[®] may be designated as ceiling limits, STELs, or 8-hr TWA limits [ACGIH 2019].

Time-weighted average (TWA): A worker's 8-hour (or up to 10-hour) time-weighted average exposure concentration that shall not be exceeded during an 8-hour (or up to 10-hour) work shift of a 40-hour week. The average concentration is weighted to take into account the duration of different exposure concentrations.

Toxicity: The degree to which a substance can cause an adverse effect on an exposed organism.

Uncertainty factors (UFs): Mathematical adjustments applied to the point of departure (POD) when developing IDLH values. The UFs for IDLH value derivation are determined by considering the study and effect used for the POD, with further modification based on the overall database.

Workplace Environmental Exposure Levels (WEELs®): Exposure levels developed by the American Industrial Hygiene Association (AIHA®) that provide guidance for protecting most workers from adverse health effects related to occupational chemical exposures, expressed as TWA or ceiling limits.

This page intentionally left blank.

Acknowledgments

This document was developed by the Division of Science Integration, Paul Schulte, Ph.D., director. R. Todd Niemeier, M.S., CIH, was the project officer and lead NIOSH author. The basis for this document was a report contracted by NIOSH and prepared by Andrew Maier, Ph.D.; Ann Parker; and Lynn Haber, Ph.D. (Toxicology Excellence for Risk Assessment [TERA]).

The following NIOSH staff are acknowledged for providing technical and editorial review of the report:

Division of Science Integration

Devin Baker, M.Ed. G. Scott Dotson, Ph.D. (former NIOSH) Sherry Fendinger, M.L.I.S. Charles L. Geraci, Ph.D. Thomas J. Lentz, Ph.D. Richard W. Niemeier, Ph.D. (retired) Pranav Rane, M.P.H. Chris Whittaker, Ph.D.

NIOSH acknowledges the following subject matter experts for their critical technical reviews:

Jeremy Brower, Ph.D., Associate Research Scientist, Lovelace Respiratory Research Institute, Albuquerque, NM

Mary E. Davis, Ph.D., Professor, Pharmacology and Toxicology, Robert C. Byrd Health Sciences Center School of Medicine, West Virginia University, Morgantown, WV

Mary A. Fox, Ph.D., Assistant Professor, Co-Director, Risk Sciences and Public Policy Institute, Department of Health Policy and Management, Bloomberg School of Public Health, Johns Hopkins University, Baltimore, MD

Randal J. Keller, Ph.D., C.I.H, D.A.B.T., Professor, Department of Occupational Safety and Health, Jesse D. Jones College of Science, Engineering, and Technology, Murray State University, Murray, KY

Mattias U. L. Öberg, Ph.D., Associate Professor, Institute of Environmental Medicine, Division of Work Environment Toxicology, Director of Risk Assessment, Swedish Toxicology Research Center (Swetox), Södertälje, Sweden

Richard B. Schlesinger, Ph.D., Fellow A.T.S., Senior Associate Dean for Academic Affairs and Research, Professor of Biology, Dyson College of Arts and Sciences, Pace University, New York, NY

George M. Woodall, Ph.D., toxicologist, National Center for Environmental Assessment, Office of Research and Development, U.S. Environmental Protection Agency, Research Triangle Park, NC

This page intentionally left blank.

1 Introduction

1.1 Overview of the IDLH Value for Bromine Trifluoride

IDLH Value: 12 ppm (67 mg/m³)

Basis for IDLH Value: Data on bromine trifluoride (BrF₃) were inadequate to directly derive an IDLH value for BrF₃. Data from studies with chlorine trifluoride (ClF_3) were used to develop an IDLH value for BrF₃ because their chemical structures, reaction mechanisms, and potencies are similar. Therefore, deriving an IDLH value based on the toxicity data for ClF₃ is appropriately health-protective. The IDLH value for ClF₃ is based on eye irritation in rats at a concentration of 480 ppm exposed for 5 minutes, which was determined to be escape impairing [Horn and Weir 1955]. The duration-adjusted 30-minute concentration is 121 ppm. An uncertainty factor (UF) of 10 was applied to account for interspecies variability (factor of 3) and human variability (factor of 3), resulting in a derived IDLH value of 12 ppm.

1.2 Purpose

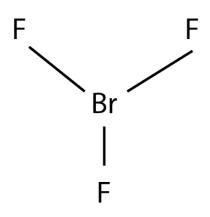
This *IDLH Value Profile* presents (1) a brief summary of technical data associated with acute inhalation exposures to ClF_3 (a surrogate for BrF_3), and (2) the rationale behind the immediately dangerous to life or health (IDLH) value for BrF_3 , using data from studies with ClF_3 as a surrogate. IDLH values are developed based on the scientific rationale and logic outlined in the *NIOSH Current Intelligence Bulletin* (*CIB*) 66: Derivation of Immediately Dangerous to Life or Health (*IDLH*) Values [NIOSH 2013].

1.3 Literature Search Strategy and Data Evaluation

As described in CIB 66, NIOSH performed in-depth literature searches to ensure that all relevant data from human and animal studies with acute exposures to the substance were identified. See Appendix A for further information about the literature search strategy and data evaluation methods.

1.4 General Substance Information

Chemical: Bromine trifluoride CAS No: 7787-71-5 Synonyms: Bromine fluoride (BrF₃)* Chemical category: Not available Reference: 'NRC [2014] Structural formula*:



Reference: *NLM [2019]

Table 1 highlights selected physiochemical properties of BrF_3 relevant to IDLH conditions. Table 2 provides alternative exposure

guidelines for BrF_3 . Table 3 summarizes the acute exposure guideline level (AEGL) values for BrF_3 .

Property	Value*
Molecular weight	136.91
Chemical formula	BrF ₃
Description	Colorless to pale yellow liquid
Odor	Not available
Odor threshold	Not available
UEL	Not applicable (nonflammable)
LEL	Not applicable (nonflammable)
Vapor pressure	2.803 g/cu cm at 25°C (77°F)
Flash point	Not applicable (nonflammable)
Ignition temperature	Not applicable (nonflammable)
Solubility	Reacts with water

Table 1. Physiochemical Properties of Bromine Trifluoride (BrF₃)

Reference: *NRC [2014]

Organization	Value
NIOSH (1994) IDLH value*	None
NIOSH REL [†]	None
OSHA PEL‡	None
	None
AIHA [®] ERPGs™¶	None
AIHA [®] WEELs ^{®**}	None

References:

*NIOSH [1994]; †NIOSH [2005]; ‡OSHA [2019]; §ACGIH [2019]; ¶AIHA [2016]; **TERA [2014]

Classification	10-min	30-min	1-hour	4-hour	8-hour	End point (Reference)
AEGL-1 (Nondisabling)	0.12 ppm 0.67 mg/m³	0.12 ppm 0.67 mg/m³	0.12 ppm 0.67 mg/m³	0.12 ppm 0.67 mg/m³	0.12 ppm 0.67 mg/m³	Set equal to AEGL-1 values for chlorine tri- fluoride [NRC 2007]
AEGL-2 (Disabling)	8.1 ppm 45 mg/m³	3.5 ppm 20 mg/m³	2 ppm 11 mg/m³	0.7 ppm 3.9 mg/m³	0.41 ppm 2.3 mg/m³	Set equal to AEGL-2 values for chlorine tri- fluoride [NRC 2007]
AEGL-3 (Lethal)	84.0 ppm 470 mg/m³	36 ppm 200 mg/m ³	21 ppm 120 mg/m³	7.3 ppm 41 mg/m³	7.3 ppm 41 mg/m³	Set equal to AEGL-3 values for chlorine tri- fluoride [NRC 2007]
Reference: NRC [2014]						

Table 3. AEGL Values for Bromine Trifluoride (BrF_3)

2 Human Data

No toxicity studies of sufficient quality were available for BrF₃ (see NIOSH 2013, section 3.4.1 for criteria to determine studies of sufficient quality). Human toxicity information is limited to a statement that concentrations of 50 ppm or more may be fatal in 30 minutes to 2 hours [Braker and Mossman 1980] but insufficient details were available to use this as the basis for an IDLH value. However, based on the chemical and physical properties, as well as data from animal studies (Section 3.0), BrF_3 is likely irritating and corrosive to skin, eyes, mucous membranes and respiratory tract. Hydrogen fluoride (HF) is the principal hydrolysis product of both BrF₃ and chlorine trifluoride (ClF_3) (3 moles of HF formed per mole of BrF₃/ClF₃). Braker and Mossman [1980] reported that the toxic effects of BrF₃ are comparable to those of ClF₃ because the toxicity of halogen fluorides is consistent with their relative reactivity. Limited data indicate that bromine pentafluoride (BrF₅) is less toxic than chlorine pentafluoride (ClF_5) (consistent with their relative reactivity); BrF₃ is similarly predicted to be less toxic than ClF₃ [Braker and Mossman 1980; NIOSH 2016a,b]. The National Research Council (NRC) used a similar strategy for development of the AEGL values for BrF₃ [NRC 2014]. In this case, the AEGL-1, AEGL-2, and AEGL-3 values for BrF₃ were based on the data for ClF₃, due to their similarities in structural activity with halogen fluorides. In addition, NRC chose to set the AEGL values for BrF₃ to the more toxic analogue, ClF₃, and did not apply any modifying factors even though BrF3 was considered less toxic [NRC 2007, 2014]. NRC indicated that the use of the ClF₃ AEGL values for BrF₃ would be "reasonably protective" [NRC 2014]. Without human and animal data for BrF₃, ClF₃ data were used to derive an IDLH value for BrF₃.

Human toxicity studies of sufficient quality for ClF_3 were limited. With moist-air, or in

the respiratory tract, ClF_3 disintegrates rapidly into HF, chlorine, chlorine dioxide, and other highly reactive compounds [Dost et al. 1974]. Consequently, the chemical is a potent irritant of mucous membranes, eyes, and skin [Teitelbaum 2001]. Studies of sufficient quality for acutely toxic concentration values or an odor threshold for humans were not identified, although Reed et al. [1966] and Deichmann and Gerarde [1969] reported without further detail that exposures of 50 ppm were lethal to humans within 30 minutes to 2 hours.

Cloyd and Murphy [1965] reported that "odorously detectable concentrations" of ClF₃ cause gasping, ocular irritation with lacrimation, cloudiness of the cornea, severe salivation, coughing and dyspnea, skin burns, and convulsions after a few minutes of exposure. Fatigue may last some time beyond the end of exposure, the corneal damage may be permanent, and skin damage may heal poorly [Cloyd and Murphy 1965]. Cloyd and Murphy [1965] also reported that concentrations of 50 ppm or more may be fatal in 15-30 minutes but did not provide further details for the basis of this reported lethal concentration. The NRC cited an incident report in which one worker was exposed for 1-2 minutes to an unknown concentration of ClF₃ [Longley et al. 1965 (as cited in NRC 1984)]. The worker complained of headache, abdominal pain, and breathing difficulty that lasted approximately 2 hours; however, no local or systemic effects were observed. The report indicated that the worker reported to work the day following exposure "with no apparent aftereffects except fatigue" [Longley et al. 1965 (as cited in NRC 1984)]. The acute symptoms of ClF₃ poisoning resemble those caused by HF [Darmer et al. 1972; MacEwen and Vernot 1970]. Also similar to HF, more severe respiratory effects (e.g., lung congestion, edema, emphysema, and diffuse hemorrhagic changes) of ClF₃ exposure may develop in a delayed fashion [HSDB 2019; MacEwen and Vernot 1970].

The weight of evidence suggests that irritative and toxic effects occur with human exposure to ClF_3 ; however, quantitative data to derive a POD for derivation of an IDLH value on human data were not identified.

3 Animal Toxicity Data

 BrF_3 is irritating and corrosive to the skin, eyes, mucous membranes, and respiratory tract. Because of the absence of empirical data for BrF_3 , this assessment uses ClF_3 as a surrogate because their chemical structures, reaction mechanisms, and potencies are similar [Braker and Mossman 1980; NRC 2014].

Multiple animal studies of sufficient quality were identified for consideration for the derivation of an IDLH value for ClF_3 . This included lethality studies [Darmer et al. 1972; Dost et al. 1967, 1974; Horn and Weir 1955; MacEwen and Vernot 1970; Vernot et al. 1977], and two nonlethal concentration studies [Horn and Weir 1955; MacEwen and Vernot 1970]. Some of the limited data on nonlethal effects of ClF_3 were from clinical observations of irritative effects in the lethality studies at concentrations or time periods less than the LC_{50} or ET_{50} derived values.

Twenty rats exposed to 5.15 ppm ClF_3 for 6 hours appeared unaffected [Horn and Weir 1955]. Two of 2 dogs exposed to this concentration for 6 hours exhibited salivation, lacrimation, rhinorrhea, and blinking of the eyes [Horn and Weir 1955]. The severity of the effects seen in dogs was not reported, so a determination of whether they were escape-impairing could not be made. In the same study, a group of 20 rats and 2 dogs were exposed to 21 ppm ClF₃ for 6 hours per day for 2 consecutive days [Horn and Weir 1955]. Rats experienced rhinorrhea and began preening shortly after a 10-minute period of exposure. The authors reported that rhinorrhea and lacrimation occurred after the first exposure period; however, no information was provided as to the severity of these effects. It was also reported that both dogs began experiencing rhinorrhea and lacrimation within 10 minutes of the exposure starting.

Additionally, it was reported that the dogs "blinked continuously at first and later kept their eyes tightly closed"; however, the time that these symptoms began was not noted [Horn and Weir 1955]. Additional data from two studies on irritative effects observed in monkeys, dogs, rats, and mice during lethal concentration studies were also identified [Horn and Weir 1955; MacEwen and Vernot 1970]. Twenty rats exposed to 480 ppm for 5 minutes experienced "considerable" lacrimation and had their eyelids closed. Some animals were observed to have swollen and raw eyelids, while other animals had considerable crusting of their eyelids [Horn and Weir 1955]. In a group of 20 rats exposed to 96 ppm for 65 minutes, one rat was in acute distress with marked gasping respiration; the other 19 rats were observed to be lying on the floor with their eyes tightly closed [Horn and Weir 1955]. In the same study, at 120 minutes, the rats were observed to have considerable exudate around the eyes, swollen and raw eyelids, crusting around the mouth and nose, and bubbling rhinorrhea [Horn and Weir 1955]. Eight rats exposed to 200 ppm for 60 minutes and 15 mice exposed to 125 ppm for 60 minutes were observed to have lacrimation, salivation, dyspnea, and rhinorrhea during the exposures [MacEwen and Vernot 1970]. Four monkeys exposed to 127 ppm for 60 minutes were observed to have signs of bronchotracheal/gastrointestinal irritation, which included sneezing, coughing, and a gagging reflex [MacEwen and Vernot 1970]. Table 4 summarizes nonlethal data reported in animal studies with 30-minute equivalent derived values for ClF₃. Information in these tables includes species of test animals, toxicological metrics (i.e., LC, BMCL, NOAEL, LOAEL), and adjusted 30-minute concentration.

Median lethal concentration (LC_{50}) and the effective median lethal time to 50% of the animal (ET_{50}) values for ClF_3 were evaluated in several animal species. MacEwen and Vernot [1970] exposed mice, rats, and monkeys to ClF₃ for 60 minutes and observed lacrimation, salivation, rhinorrhea, and dyspnea that, within a few hours after exposure, turned into bloody discharges if the animals survived. Monkeys also showed signs typical of bronchial and gastrointestinal irritation. Death occurred with delays as long as 36 hours after exposure. Upon death, massive alveolar and interstitial hemorrhage were noted. Near-fatal concentrations resulted in concentration-dependent pulmonary congestion, edema, emphysema, and hemorrhage. The 60-minute LC_{50} values were 178 ppm for mice, 299 ppm for rats, and 230 ppm for monkeys (also reported in Darmer et al. [1972]).

Horn and Weir [1955] exposed rats to two concentrations of ClF₃ and determined the ET_{50} . In rats, the ET_{50} at 480 ppm was 40 minutes (all dead within 70 minutes), at 96 ppm it was 3.7 hours (observation time after the end of the 4.5-hour exposure to 96 ppm was not stated). Clinical signs appeared within minutes of exposure and included increased activity, nasal flow and salivation, respiratory difficulty, eye irritation, and convulsions and coma shortly before death. Dost et al. [1967, 1974] reported that ClF₃ caused severe inflammation in all exposed tissues, lacrimation, and shallow breathing in male rats. High concentrations made hair appear "singed," caused skin burns, and produced corneal ulceration. These authors also observed that rats surviving ClF₃ exposure for 4 hours did not eat for several days thereafter. Time to death was tested in the presence of 400 and 800 ppm ClF_3 ; all animals died within 45-90minutes of exposure to 800 ppm for 15 minutes; at longer exposure times, up to 30 minutes, the earliest deaths occurred within 20 minutes, but some animals survived as long as 160 minutes. At 400 ppm, death occurred after 55–140 minutes with \geq 30 minutes exposure, but no deaths were observed at ≤ 25 minutes exposure. NRC [2007, 2014] provided an estimated 1-hour LC_{50} value of 222

ppm based on these data but indicated that this value may be an underestimate because post-exposure observations were not completed [NRC 2007, 2014]. Table 5 summarizes the LC data identified in animal studies and provides 30-minute equivalent derived values for ClF_3 . Information in this table includes species of test animals, toxicological metrics (i.e., LC, BMCL, NOAEL, LOAEL), and adjusted 30-minute concentration.

The weight of evidence suggests that ClF₃ irritates the respiratory and ocular systems in the four animal species tested, and the effects occurred at concentrations as low as 5.15 ppm. Lethal concentrations were also available for three animal species but were higher than the irritative effects and were excluded from consideration as a POD for the IDLH value. The most sensitive species appears to be rats, in which irritative ocular effects, determined to be escape-impairing, began to occur at 480 ppm at a 5-minute exposure period, resulting in a 30-minute time-adjusted concentration of 121 ppm [Horn and Weir 1955]. This endpoint was selected as the POD for the derivation of the IDLH value. Similar irritative effects were observed at similar 30-minute time-adjusted concentrations in multiple animal species and ranged from 142–341 ppm. One study identified irritative effects at a lower concentration than the chosen POD [Horn and Weir 1955]. In this study, dogs exposed at concentrations of 5.15 ppm over a 360-minute exposure period experienced irritation, lacrimation, rhinorrhea, coughing and sneezing; however, the severity of these effects was not provided. The authors concluded that at the end of the first exposure period the dogs were not markedly affected. This concentration was not considered to be escape-impairing, so it was not considered as the POD for the derivation of the IDLH value. See Section 4 for information regarding the derivation of the IDLH for ClF₃, including additional information on the study selected as the POD and the uncertainty factors (UFs) applied to derive a final IDLH value for ClF_{3} , as the surrogate for BrF_{3} .

Horn and Weir Dog Sali [1955] Shi Rhi kep		number or animals	Dose (ppm)	Time (min)	30-min Concentration [†] (ppm)
Rhi Kep	Salivation, lacrimation, rhinorrhea, severe coughing, sneezing.	0	5.15	360‡	35
	Rhinorrhea, lacrimation, continuous eye blinking, kept eyes closed tightly	0	21	360 [§]	142
nd Weir Rat	None reported	20	5.15	360‡	35
[1955] Rhi	Rhinorrhea and lacrimation	20	21	360 [§]	142
Cor anii had	Considerable lacrimation and eye lids closed, some animals had swollen and raw eye lids, some animals had considerable crusting of eyelids	20	480	5 ¹	121
1 0 19 0 10 0 0 0 0	 of 20 rats in acute distress with marked gasping respiration, lacrimation, salivation, and rhinorrhea. of 20 rats lying on cage floor with eyes tightly closed 	20	90	65 65 8	174
Cor and bub	Considerable exudate around eyes with swollen and raw eyelids, crusting around mouth and nose, bubbling rhinorrhea	20	96	120**	279
MacEwen and Monkey Bro Vernot [1970] ing,	Bronchotracheal/gastrointestinal irritation (sneez- ing, coughing, and gagging reflex)	4	127	60††	217
MacEwen and Rat Lac Vernot [1970] rhir	Lacrimation, salivation, dyspnea, rhinorrhea during exposures.	Ø	200	60 ^{††}	341
MacEwen and Mouse Lac Vernot [1970] rhir	Lacrimation, salivation, dyspnea, rhinorrhea during exposures.	15	125	60 ^{††}	213

Table 4. Nonlethal Concentration Data for Chlorine Trifluoride (CIF_3)*

aujusteu uuration 5 n = time scaling value; NRC [2007] provided an empirically estimated n of 1.3 for all time-scaling. Additional information on the calculation trations can be found in NIOSH [2013].

⁺The exposure time and symptoms noted were from the first daily exposure of a 6 week, 6-hour-per-day, 5-day-per-week inhalation study.

^sThe exposure time and symptoms noted were from the first daily exposure of a 2 day, 6-hour-per-day inhalation study.

^TThe exposure time and symptoms noted were from an ET₅₀ acute inhalation study at a concentration of 480 ppm. ^{**}The exposure time and symptoms noted were from an ET₅₀ acute inhalation study at a concentration of 96 ppm. ^{††}The exposure time and symptoms noted were from an LC₅₀ acute inhalation study.

		Number					Adjusted
Reference	Species	of animals	Dose(s) (ppm)	LC ₅₀ (ppm)	ET ₅₀ (ppm)	Time (min)	30-min Concentration [†] (ppm)
Darmer et al. [1972]; MacEwen and Vernot [1970]	Monkey	5/dose	127, 150, 200, 300, 400	230	1	60	392
Darmer et al. [1972]; MacEwen and Vernot [1970]; Vernot et al. [1977]	Rat	8/dose	200, 400	299	1	60	510
Darmer et al. [1972]; MacEwen and Vernot [1970]; Vernot et al. [1977]	Mouse	15/dose	125, 150, 175, 200, 400	178	I	60	303
Horn and Weir [1955]	Rat	20/dose	480	ı	480	40	599
			96	I	96	222	448
Dost et al. [1967, 1974]	Rat	34	400	222 [‡]		20-40	0 E C
		46	800			10-30	010

Table 5. Lethal Concentration Data for Chlorine Trifluoride (CIF $_{a})^{*}$

n = time scaling value; NRC [2007] provided an empirically estimated n of 1.3 for all time-scaling. Additional information on the calculation of duration adjusted concen-trations can be found in NIOSH [2013]. #Estimated value based on NRC [2007] extrapolation of Dost et al. [1974] data.

4 Summary

Inadequate toxicity data for BrF₃ were available to directly derive an IDLH for BrF₃. The data on ClF₃ were used to derive an IDLH for BrF₃ because their structures, reaction mechanisms, and potencies are similar. Additionally, BrF₂ shares a common hydrolysis product (HF) with ClF₃. HF is known to be highly corrosive and irritating to the skin, eyes, and respiratory tract. Therefore, deriving an IDLH value based on the toxicity data for ClF₃ is appropriately health-protective [NIOSH 2019]. Braker and Mossman [1980] reported that the toxic effects of BrF₃ are comparable to those of ClF₃, because the toxicity of halogen fluorides is consistent with their relative reactivity. In the absence of such data for BrF₃, ClF₃ was used as a surrogate to derive an IDLH value for BrF₃. The NRC used a similar strategy for development of the AEGL values for BrF₂ [NRC 2014].

Several acute toxicity studies with exposure to ClF₃ were identified. After adjustment to a 30-minute exposure duration, nonlethal concentrations corresponding to escape-impairing health effects ranged from 121 to 341 ppm. These concentrations were time-concentration adjusted to 30-minute exposure durations using the ten Berge methodology with n = 1.3 [ten Berge et al. 1986; NRC 2007]. It was determined appropriate to use a time-concentration factor for the irritative endpoint because these tissue damage effects caused by a direct-contact irritant appeared to be along a continuum [NRC 2007]. In one lower concentration (30-minute equivalent of 35 ppm), irritative symptoms were observed in dogs but insufficient information was presented on the severity of the effects, so a determination whether that concentration was escape-impairing could not be made [Horn and Weir 1955]. Some of the exposure and symptom data were collected during lethal concentration studies; however, the symptoms reported were useful in identifying irritative effects that may be escape impairing. LC₅₀ values in experimental animals range from 303 to 599 ppm [Darmer

et al. 1972; MacEwen and Vernot 1970; Horn and Weir 1955; Dost et al. 1974, 1967]. Because these values were higher than the irritative effects, their use would require the application of additional UFs, and thus they were excluded from consideration as a POD for the IDLH value. Additionally, because the irritative effects for ClF₂ appear at lower concentrations in the cascade of adverse effects, the irritation endpoint was chosen as the POD. However, if the lethality endpoint was chosen as the POD, the final IDLH values would have been within a factor of 4 after applying appropriate UFs (UF = 30–100). The most sensitive endpoint identified was in 20 rats exposed to 480 ppm as part of a lethal concentration study. As part of this study, it was observed that after 5 minutes of exposure, the rats experienced considerable lacrimation; kept their eyelids closed; and had swollen, raw, and crusting eyelids [Horn and Weir 1955]. These effects were determined by NIOSH to be severe ocular irritation that would be escape-impairing and thus this was used as the POD for the IDLH derivation. An adjusted 30-minute concentration was calculated from 5 minutes of exposure with a final POD of 121 ppm. A composite UF of 10 was applied to account for extrapolation from interspecies variability (animal to human), characterized by a factor of 3; and human variability, characterized by a factor of 3, resulting in a POD of 121 ppm divided by an UF of 10 or an IDLH value of 12 ppm. The justification for the underlying component UFs is as follows:

• Interspecies UF (=3)

This value was selected because rats were the most sensitive animal species tested when comparing the 30-minute time-adjusted concentrations, and interspecies differences are not as pronounced for direct-acting irritative effects as with other endpoints.

• Intraspecies UF (=3)

This value was selected because the difference in sensitivity to irritants among workers is typically less than other endpoints.

5 References

ACGIH[®] (American Conference of Governmental Industrial Hygienists) [2019]. Annual TLVs[®] (threshold limit values) and BEIs[®] (biological exposure indices) booklet. Cincinnati, OH: ACGIH Signature Publications.

ACGIH (American Conference of Governmental Industrial Hygienists) [2020]. Documentation of the threshold limit values and biological exposure indices, 7th edition – 2020 supplement. Cincinnati, OH: American Conference of Governmental Industrial Hygienists, https://www. acgih.org/forms/store/ProductFormPublic/documentation-of-the-threshold-limit-values-and-biological-exposure-indices-7th-ed-2020-supplement. Date accessed: January 8, 2020.

AIHA[™] (American Industrial Hygiene Association) [2016]. AIHA Emergency Response Planning (ERP) Committee procedures and responsibilities. Fairfax, VA: American Industrial Hygiene Association.

Braker W, Mossman AL [1980]. Bromine pentafluoride. In: Matheson gas data book. 6th ed. Lyndhurst, NJ: Matheson Gas Products.

Cloyd DR, Murphy WJ [1965]. Handling hazardous materials. Washington, DC: National Aeronautics and Space Administration, NASA SP-5032.

Darmer KI, Haun CC, MacEwen JD [1972]. The acute inhalation toxicity of chlorine penta-fluoride. Am Ind Hyg Assoc J 33:661–668.

Dost FN, Reed DJ, Smith VN, Wang CH [1967]. Metabolism and pharmacology of inorganic and fluorine containing compounds. Wright-Patterson Air Force Base, OH: Aerospace Medical Research Laboratories, Aerospace Medical Division, Air Force Systems Command, AMRL-TR-67-224.

Dost FN, Reed DJ, Smith VN, Wang CH [1974]. Toxic properties of chlorine trifluoride. Toxicol Appl Pharmacol 27:527–536.

EPA [2019]. Integrated Risk Information System (IRIS): glossary, https://ofmpub.epa.gov/

sor_internet/registry/termreg/searchandretrieve/glossariesandkeywordlists/search.do?details=&glossaryName=IRIS%20Glossary#formTop. Accessed June 11, 2019.

Horn HJ, Weir RJ [1955]. Inhalation toxicity of chloride trifluoride. I. Acute and subacute toxicity. AMA Arch Ind Health *12*:515–521.

HSDB [2019]. Hazardous Substances Data Bank, HSDB. CAS No. 7790-91-2, https://haz-map. com/. Date accessed: June 11, 2019.

Longley MY, Pierce JF, Griesemer EC [1965]. A toxic hazard study of selected missile propellants. AMRL-TR-65-99. Wright-Patterson Air Force Base, OH: Aerospace Medical Research Laboratories.

MacEwen JD, Vernot EH [1970]. Toxic hazards research unit. Annual technical report: 1970. Wright-Patterson Air Force Base, OH: Aerospace Medical Research Laboratories, Aerospace Medical Division, Air Force Systems Command, AMRL-TR-70-77.

NAS (National Academy of Sciences) [2000]. Emergency and continuous exposure limits for selected airborne contaminants. Vol. 2. National Academy of Sciences, National Research Council (NRC), Committee on Toxicology, Board of Toxicology and Environmental Health Hazards, Commission on Life Sciences, National Research Council. Washington, DC: National Academy Press.

NIOSH (National Institute for Occupational Safety and Health) [1994]. Documentation for immediately dangerous to life or health concentrations (IDLHs): Chlorine trifluoride, https://www.cdc.gov/niosh/idlh/7790912.html. Date accessed: January 10, 2018.

NIOSH [2004]. NIOSH respirator selection logic. By Bollinger N. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2005-100. NIOSH [2005]. NIOSH pocket guide to chemical hazards. Barsan ME, ed. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2005-149.

NIOSH [2013]. Current intelligence bulletin 66: derivation of immediately dangerous to life or health (IDLH) values. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2014-100.

NIOSH [2016a]. Immediately dangerous to life or health (IDLH) value profile: bromine pentafluoride. By Dotson GS, Maier A, Parker A, Haber L. Cincinnati, OH: US Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication 2017-106.

NIOSH [2016b]. Immediately dangerous to life or health (IDLH) value profile: chlorine pentafluoride. By Dotson GS, Maier A, Parker A, Haber L. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication 2016-170.

NIOSH [2020]. Immediately dangerous to life or health (IDLH) value profile: chlorine trifluoride. By Niemeier RT. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2020-124.

NLM (National Library of Medicine) [2019]. ChemIDplus lite, https://chem.sis.nlm.nih. gov/chemidplus/chemidlite.jsp. Date accessed: June 11, 2019.

NRC (National Research Council) [1984]. Emergency and continuous exposure limits for selected airborne contaminants. Vol. 2. Washington, DC: The National Academies Press, https://doi. org/10.17226/690. NRC [2001]. Standing operating procedures for developing acute exposure guideline levels for hazardous chemicals. Washington, DC: The National Academies Press, https://doi. org/10.17226/10122.

NRC [2007]. Chlorine trifluoride. In: Acute exposure guideline levels for selected airborne chemicals. Vol. 5. Washington, DC: The National Academies Press, pp. 53–91, https://doi.org/10.17226/11774.

NRC [2014]. Bromine trifluoride. In: Acute exposure guideline levels (AEGLs) for selected airborne chemicals. Vol. 18. Washington, DC: The National Academies Press, pp. 13–36, https://doi.org/10.17226/18941.

OSHA (Occupational Safety and Health Administration) [2019]. Occupational safety and health standards. 29 CFR 1910. Subpart Z—toxic and hazardous substances. Washington, DC, https://www.osha.gov/pls/oshaweb/owa-disp.show_document?p_table=standards&p_id=9992. Date accessed: June 11, 2019.

Reed DJ, Dost FN, Wang CH [1966]. Inorganic fluoride propellant oxidizers. Vol. I. Their effects upon seed germination and plant growth. Wright-Patterson Air Force Base, OH: Aerospace Medical Research Laboratories, Aerospace Medical Division, Air Force Systems Command.

Teitelbaum DT [2001]. The halogens. In: Bingham E, Cohrssen B, Powell CH, eds. Patty's toxicology. Vol. 3. New York: John Wiley & Sons, Inc., pp. 731–825.

ten Berge WF, Zwart A, Appelman LM [1986]. Concentration-time mortality response relationship of irritant and systematically acting vapors and gases. J Haz Mat *13*:301–309.

TERA (Toxicology Excellence for Risk Assessment) [2014]. Occupational alliance for risk science-workplace environmental exposure levels (WEEL)[™], https://www.tera.org/OARS/WEELs.pdf. Date accessed: June 11, 2019.

Appendix A—Literature Search Strategy and Data Evaluation

As described in Current Intelligence Bulletin 66 [NIOSH 2013], NIOSH performed indepth literature searches to ensure that all relevant data from human and animal studies with acute exposures to the substance were identified. This search was conducted in July 2019 and included available literature to July 2019. This process included reviewing toxicity summaries and acute exposure guidelines during an initial screening approach, including the AEGL documentation [NRC 2014].

Studies were identified from these summaries and additional database queries were conducted in the databases listed in Table A-1. Search terms used in these databases are provided in Table A-2. For BrF_3 , the chemical identifiers used in the database searches were "bromine trifluoride," "bromotrifluoride," "bromine fluoride," and "CAS Registry Number: 7787-71-5."

Inadequate toxicity data were available for BrF_3 . The data on ClF_3 were used to derive an IDLH for BrF_3 because their structures, reaction mechanisms, and potencies are assumed to be similar. For ClF_3 , this process included reviewing toxicity summaries and acute exposure guidelines during an initial screening approach, including the AEGL documentation [NRC 2007], ACGIH[®] TLV[®] documentation [ACGIH 2020], NIOSH 1994 IDLH documentation [NIOSH 1994], and the Emergency and Continuous Exposure Limits for Selected Airborne Contaminants documentation [NRC 1984]. Studies were identified from these summaries and additional database queries were conducted in the databases listed in Table A-1. Search terms used in these databases are provided in Table A-2. For CIF₃, the chemical identifiers used in the database searches were "chlorine trifluoride," "chlorotrifluoride," "chlorine fluoride," and "CAS Registry Number: 7790-91-2."

For ClF₃, the in-depth literature search was conducted through June 2019. The information that was identified in the in-depth literature search was evaluated with general considerations that included description of studies (i.e., species, study protocol, exposure concentration, and duration), health endpoint evaluated, and critical effect levels (e.g., NOAELs, LOAELs, and LC_{50} values). Eighteen references were identified in the literature search and 10 of these were cited in the IDLH document. These references included both peer-reviewed literature and scientific reports. The eight references that were not cited did not provide primary data on ClF_3 .

Database and databases	Link
Centers for Disease Control and Preven- tion (CDC)/ Agency for Toxic Substance and Disease Registry (ATSDR) ToxProfiles	https://www.atsdr.cdc.gov/toxprofiles/index.asp
ChemIDplus	https://chem.sis.nlm.nih.gov/chemidplus/chemid- lite.jsp
EU, European INventory of Existing Com- mercial chemical Substances (EINECS)	https://ihcp.jrc.ec.europa.eu/our_labs/predictive_ toxicology/information-sources/ec_inventory
EMBASE	https://www.embase.com/
Hazardous Substances Data Bank (HSDB)	https://www.nlm.nih.gov/databases/download/ hsdb.html
Haz-map	https://haz-map.com/
National Library of Medicine (NLB)	https://pubchem.ncbi.nlm.nih.gov/
International Agency for Research on Cancer (IARC)	https://www.iarc.fr/
Environmental Protection Agency (EPA) Toxic Substance Control Act (TSCA) Sec- tion 8(e) Notices	https://www.epa.gov/oppt/tsca8e/index.html
World Health Organization (WHO)/IPCS International Chemical Safety Card (ICSC)	https://www.ilo.org/safework/langen/index.htm
International Toxicity Estimates for Risk (ITER)	https://www.iter.tera.org/
New Jersey Hazardous Substance Fact Sheets (NJ-HSFS)	https://web.doh.state.nj.us/rtkhsfs/indexfs.aspx
NIOSHTIC-2	https://www2a.cdc.gov/nioshtic-2/default.asp
NLM, PUBMED	https://www.ncbi.nlm.nih.gov/pubmed/
National Institute for Occupational Safety and Health (NIOSH)/Registry of Toxic Ef- fects of Chemical Substances (RTECS)	https://web.doh.state.nj.us/rtkhsfs/indexfs.aspx
NLM, Toxicology Literature Online (TOXLINE)	https://www.ncbi.nlm.nih.gov/pubmed?term=- tox%20%5Bsubset%5D%20AND%20
Scopus	https://www.scopus.com/search/form.uri?dis- play=basic

Table A-2. Literature search terms

	Search terms	
Acute	Fatal	Threshold
Inhalation	Fatal	Case study
Lethal	Irritation	Poisoning
Lethal concentration	Respiratory	Chemical identifiers
LC	RD	

References

ACGIH (American Conference of Governmental Industrial Hygienists) [2020]. Documentation of the threshold limit values and biological exposure indices, 7th edition – 2020 supplement. Cincinnati, OH: American Conference of Governmental Industrial Hygienists, https://www. acgih.org/forms/store/ProductFormPublic/documentation-of-the-threshold-limit-values-and-biological-exposure-indices-7th-ed-2020-supplement. Date accessed: January 8, 2020.

NIOSH [2013]. Current intelligence bulletin 66: derivation of immediately dangerous to life or health (IDLH) values. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2014-100. NIOSH [1994]. Documentation for immediately dangerous to life or health concentrations (IDLHs): chlorine trifluoride, https://www.cdc. gov/niosh/idlh/7790912.html. Date accessed: June 11, 2019.

NRC (National Research Council) [1984]. Emergency and continuous exposure limits for selected airborne contaminants. Vol. 2. Washington, DC: The National Academies Press, https://doi.org/10.17226/690.

NRC [2007]. Chlorine trifluoride. In: Acute exposure guideline levels for selected airborne chemicals. Vol. 5. Washington, DC: The National Academies Press, pp. 53–91, https://doi.org/10.17226/11774.

NRC [2014]. Bromine trifluoride. In: Acute exposure guideline levels (AEGLs) for selected airborne chemicals. Vol. 18. Washington, DC: The National Academies Press, pp. 13– 36, https://doi.org/10.17226/18941. This page intentionally left blank.



Promoting productive workplaces through safety and health research

DHHS (NIOSH) Publication No. 2020–123 DOI: https://doi.org/10.26616/NIOSHPUB2020123