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Assessing the accuracy of commercially available gas sensors for the measurement of ambient ozone and nitrogen dioxide

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Abstract

The objective of the National Institute for Occupational Safety and Health (NIOSH) accuracy criterion is to ensure that measurements from monitoring devices are within $\pm 25\%$ of the true concentration of the analyte with 95% certainty. To determine whether NO₂ and O₃ sensors meet this criterion, three commercially available units (Cairclip O₃/NO₂, Aeroqual NO₂, and Aeroqual O₃ sensors) were co-located three times with validated instruments (NO_x chemiluminescence [NO_{2mon}] and photometric O₃ analyzers [O_{3mon}]) at an outdoor monitoring station. As cofactors of sensor performance such as temperature (T) and relative humidity (RH) potentially influence the response of NO₂ and O₃ sensors, corrections for cofactors were made by using T, RH, and the sensor measurements to predict measurements made by NO_{2mon} and O_{3mon} during the first co-location period (training dataset). The developed models were tested in the merged data obtained from the second and third co-location periods (testing dataset). In the training and testing datasets, the mean NO₂ as measured by NO_{2mon} was 4.6 ppb (range = 0.4 – 35 ppb) and 9.4 ppb (range = 1 – 37 ppb), respectively. The mean O₃ in the training and testing datasets as measured by O_{3mon} was 38.8 ppb (range = 1 – 65 ppb) and 35.7 ppb (range = 1 – 61 ppb), respectively. None of the sensor measurements in the training dataset were within the NIOSH accuracy criterion (mean error $\leq 25\%$). After correcting for cofactors of sensor performance, the accuracy of the Cairclip O₃/NO₂ and the Aeroqual O₃ sensors considerably improved when tested with the testing dataset (mean error = -1% and 14% , respectively). However, the Aeroqual NO₂ sensor had an error that was not within $\pm 25\%$. Raw measurements from the tested sensors may be unsuitable for assessing workers' exposure to NO₂ and O₃. Corrections for cofactors of Cairclip O₃/NO₂ and Aeroqual O₃ sensor performance are required for more accurate occupational exposure assessment.

Keywords

Aeroqual; Cairclip; direct-reading instruments; low-cost sensors

INTRODUCTION

Nitrogen dioxide (NO₂) and ground-level ozone (O₃) are common air pollutants associated with adverse respiratory health effects.^(1–8) Due to the emission of NO₂ from vehicular exhausts,⁽⁹⁾ NO₂ is regarded as a traffic-related air pollutant, whereas O₃ is formed through photochemical reactions involving NO_x (nitrogen oxides) and volatile organic compounds,^(10, 11) which are emitted from vehicular exhausts. Consequently, traffic control workers may have elevated levels of exposure to NO₂ and O₃, which may be associated with lower lung function and increased respiratory symptoms.^(12–14) As a result of emissions from the exhausts of internal combustion engines, other outdoor workers such as toll booth workers,⁽¹⁵⁾ commercial drivers,^(13, 16) petrol-pump workers,⁽¹⁷⁾ air cargo handlers,⁽¹⁸⁾ tunnel construction workers⁽¹⁹⁾ and street cleaners^(20, 21) may also be exposed to higher levels of NO₂ and O₃ than the general population. The lowest occupational exposure limit (OEL) for NO₂ and O₃ are the threshold limit values (TLVs): 200 ppb for NO₂,⁽²²⁾ and, 50 – 200 ppb for O₃ (depending on working conditions).⁽²³⁾ In most typical outdoor scenarios, outdoor workers in countries that have enforceable regulatory ambient air quality standards are likely to have occupational exposure to outdoor NO₂ and O₃ below OEL. However, some indoor workers in developed countries and indoor/outdoor workers in developing countries can have NO₂ and O₃ exposures that exceed OELs. Examples are indoor ice skating workers in rinks that use petroleum-based fuels in the resurfacers to maintain ice^(24, 25) and accidental release of O₃ in pulp mills.⁽²⁶⁾ Thus, reliable NO₂ and O₃ monitoring devices are needed for occupational exposure assessment.

In contrast to traditional passive samplers for NO₂ and O₃,^(27, 28) electronic gas sensors are relatively new direct reading instruments that are used for assessing exposure to NO₂ and O₃. They are sometimes referred to as “low-cost sensors” because of their relatively affordable price (< \$1500). Researchers have questioned the accuracy of NO₂ and O₃ sensors, mainly because cofactors such as reactive pollutants, temperature, and relative humidity can influence the response of the sensors to NO₂ and O₃ concentrations.^(29, 30) Due to the concern about the quality of measurements from NO₂ and O₃ sensors, scientists have investigated their accuracy by co-locating them with validated instruments and utilizing Pearson correlation coefficients (r) and coefficients of determination (R²) to assess the quality of the data.^(31–33) To improve the correlation, the impact of T, RH, and reactive pollutants on sensor performance have been modeled. Applying these models, Cross et al.⁽³²⁾ and Zimmerman et al.⁽³⁴⁾ reported an increase in the correlation between corrected sensor measurements and measurements from validated instruments. Nevertheless, computed r and R² based on normality assumption may be inappropriate when the distribution of the analyte is not normally distributed.⁽³⁵⁾ According to scientists at the National Institute for Occupational Safety and Health (NIOSH), direct reading instruments for gaseous exposure assessment should measure within ± 25% of the true concentration of the target analyte with 95% certainty.⁽³⁶⁾ This assessment corresponds to the 95% confidence interval (95% CI) accuracy level. To the best of our knowledge, researchers have not demonstrated that NO₂ and O₃ gas sensors meet this accuracy criterion in the field.

The United States Environmental Protection Agency (EPA) performs regulatory actions for ambient NO₂ and O₃ concentrations through Ambient Air Quality Standards.^(37, 38) EPA

judges the accuracy of NO₂ and O₃ measuring devices by their performance in detecting NO₂ and O₃ after known concentrations are released onto the measuring devices. The EPA acceptable measurement of ambient NO₂ has a threshold of 15% coefficient of variation and bias, while that of O₃ has a threshold of 7% coefficient of variation and bias.⁽³⁹⁾ The EPA has a list of designated reference and equivalent instruments that can be used to achieve EPA acceptable measurements of ambient NO₂ and O₃.⁽⁴⁰⁾ This list includes NO_x chemiluminescence analyzers as a method for measuring ambient NO₂⁽⁴⁰⁾ and photometric O₃ analyzers as a method for measuring ambient O₃.^(40, 41) However, NO₂ and O₃ sensors are not included in the list as reference/equivalent instruments. A non-exhaustive list of commercially available NO₂ and O₃ sensors includes: Cairclip O₃/NO₂ sensor,⁽⁴²⁾ Aeroqual GSE NO₂ sensor,⁽⁴³⁾ Aeroqual GSS O₃ sensor,⁽⁴⁴⁾ MSA NO₂ detector,⁽⁴⁵⁾ Drager NO₂ sensor,⁽⁴⁶⁾ and Drager O₃ sensor.⁽⁴⁶⁾ The Aeroqual O₃ sensor is a metal-oxide-semiconductor ozone sensor,⁽⁴⁴⁾ whereas the other sensors are electrochemical sensors. Documentation from the sensor indicates that the Cairclip O₃/NO₂ sensor is capable of measuring the sum of NO₂ and O₃,⁽⁴²⁾ whereas the other sensors named are capable of measuring only their target analyte.^(43–46) Based on the theory of operation presented in the scientific literature, electrochemical sensors contain electrodes which detect the current produced from a target gas undergoing a reaction in the sensor.^(47, 48) Metal-oxide-semiconductor gas sensors have semiconductors that detect a change in electrical resistance caused by the reaction of a sampled target gas with reducing or oxidizing gases present on the surface of the semiconductor.^(49, 50) There is an inherent lack of sensitivity of target gases in metal-oxide semiconductor sensors.⁽⁵¹⁾

As gas sensors are used for the same purpose as reference/equivalent instruments (i.e., to measure air pollutants relevant to health), this study examined the accuracy of three commercial NO₂ and O₃ gas sensors in the field to validate their use for occupational exposure assessment using the NIOSH accuracy criterion. Additionally, the study examined the correlation of measurements made with the gas sensors and reference/equivalent instruments.

METHODS

Investigating the Accuracy of NO₂ and O₃ Sensors

One of each sensor [the NO₂ sensor (GSE, s500, 0–1 ppm, Aeroqual), O₃ sensor (GSS, s500, 0–0.15 ppm, Aeroqual) and the O₃+NO₂ sensor (O₃/NO₂, Cairclip, Cairpol)] was co-located outdoors three times side-by-side with the Southwest Ohio Air Quality Agency monitoring station (250 William Howard Taft Rd, Cincinnati, OH 45219). All the sensors were factory calibrated by the manufacturers, and zero calibration was performed on the Aeroqual sensors by the study team. The factory calibration of the Cairclip O₃/NO₂ sensor is sufficient for one year provided that its operational conditions are adhered to.⁽⁴²⁾ Prior to the commencement of the current study, the sensors had been used for approximately five months since the receipt from the manufacturers. They were within the recommended one-year factory calibration cycle given by the manufacturers. The first co-locating period was between 3 pm on 07/19/2017 to 2 pm on 07/24/2017, and the second and third co-locating periods were between 3 pm on 08/30/2017 to 2 pm on 09/01/2017 and 3 pm on 10/17/2017

to 2 pm on 10/19/2017, respectively. The sensors were deployed to compare the readings obtained from them to those of the EPA reference/equivalent instruments at the monitoring station (Figure S1). The monitoring station uses a NO_x chemiluminescence analyzer (Model T200, Teledyne) [NO_{2mon}] and a photometric O₃ analyzer (Model 400E, Teledyne) [O_{3mon}]. The recommended operating conditions of the measuring devices are -20 – 40°C and 10 – 90% RH for the Cairclip O₃/NO₂ sensor, 0 – 40°C and 15 – 90% RH for the Aeroqual NO₂ sensor, and 0 – 40°C and 10 – 90% RH for the Aeroqual O₃ sensor, 5 – 40°C and 0 – 95% RH for the NO_x chemiluminescence analyzer, 5 – 40°C and 0 – 90% RH for the photometric O₃ analyzer.^(43, 44, 52–54) Data from the NO_x chemiluminescence analyzer and photometric O₃ analyzer were added together to obtain “reference O₃+NO₂”. The limits of detection (LOD) of the measuring devices were obtained from their manuals (20 ppb for the Cairclip O₃/NO₂ sensor, 5 ppb for the Aeroqual NO₂ sensor, 1 ppb for the O₃ sensor, 0.4 ppb for the NO_x chemiluminescence analyzer and 0.6 ppb for the photometric O₃ analyzer).^(43, 44, 52–54)

Temperature (T) and relative humidity (RH) were simultaneously measured with a temperature/relative humidity monitor (O83E, Met One Instruments, Inc.). The time and date on all devices were synchronized and were set to record measurements every minute. For quality assurance, data from the reference/equivalent instruments were reviewed by the operators of the monitoring station (Southwest Ohio Air Quality Agency). At the end of the sampling period, hourly averages of NO₂, O₃, T, and RH were provided by the operators of the monitoring station.

All the data logged per minute by the sensors (including data < LOD) were downloaded to a computer using the manufacturers’ proprietary software, and hourly averages of NO₂ and O₃ levels measured by the sensors were calculated. Data obtained from the reference/equivalent instruments at the monitoring station were regarded as the “gold standards” providing the true concentration of ambient NO₂ and O₃. Data from all measurements were imported into R Studio⁽⁵⁵⁾ for data analysis. Only hourly averaged measurements above the LOD of the sensors and reference instruments were used in this study. First, accuracy was determined by estimating the error associated with measurements of the NO₂ and O₃ sensors. This was achieved by creating variables that contained the percentage difference between the measurements made by the sensors and measurements from the reference instruments (equation 1).

$$Error = \left(\frac{Sensor\ measurements_i - Reference\ instrument\ measurements_i}{Reference\ instrument\ measurements_i} \right) \times 100 \quad (1)$$

where i = individual observation in the dataset.

Second, the mean error was calculated and a bias-corrected and accelerated (BCa) bootstrap method was used to calculate 95% CI for the mean error.^(56–59) Bootstrap was employed for the calculation of 95% CI because the created error variables were non-normal. Accuracy was defined as having a 95% CI of mean error (lower 95% CI of mean error < true mean error < upper 95% CI of mean error) within - 25 to +25%.

The correlation of measurements from the sensors and reference instruments were also calculated in order to make results from the current study comparable to results of existing studies. The following tests for accuracy and correlation of the sensor measurements compared to the O₃ reference-equivalent instrument (O_{3mon}), NO₂ reference instrument (NO_{2mon}) and (O₃+NO₂)_{mon} were carried out:

- Aeroqual O₃ sensor and O_{3mon}
- Aeroqual NO₂ sensor and NO_{2mon}
- Cairclip O₃/NO₂ sensor and (O₃+NO₂)_{mon}

Developing Corrective Models

Data from the first co-locating period was utilized to develop regression models to correct for measured cofactors of sensor performance that potentially interfere with the accuracy of the sensors. The data collected during this period is described as the “Training Dataset.” To develop a model to correct for measured cofactors of the Aeroqual O₃ sensor performance, and to correct for the inherent error in the Aeroqual O₃ sensor, measurements of T, RH, NO_{2mon}, and the Aeroqual O₃ sensor were used to predict O_{3mon} (equation S1). Due to the non-linear response of metal-oxide semiconductor sensors to the target gas and cofactors of sensor response,⁽⁵¹⁾ quadratic terms were added to the independent variables. Variable selection was performed with a stepwise regression model using the minimum Akaike Information Criterion (AIC) for model selection. To develop a model to correct the effect of measured cofactors of the Aeroqual NO₂ sensor performance, and to correct for the inherent error in the Aeroqual NO₂ sensor, measurements of T, RH, O_{3mon}, and the Aeroqual NO₂ sensor were used to predict NO_{2mon} (equation S2). Lastly, to develop a model to correct the effect of measured cofactors of the Cairclip O₃/NO₂ sensor performance, and to correct for the inherent error in the Cairclip O₃/NO₂ sensor, measurements of T, RH, and the Cairclip O₃/NO₂ sensor were used to predict (O₃+NO₂)_{mon} (equation S3). After developing the regression models as described, the independent variables that were not significant (P > 0.05) were removed from the models, and the final models were used to correct the raw sensor data in the subsequent field trial.

Testing the Models Developed

The sensors were co-located at the monitoring station twice to test whether the accuracy of the sensor measurements could be improved by utilizing the regression coefficients obtained from the final models. The data obtained to test the models (testing dataset) were collected from the second and third co-locating periods. These data were merged together to form one testing dataset. Measurements from the testing dataset were multiplied by the corresponding coefficients obtained from the final models.

Errors of the corrected sensor measurements were calculated as described in Equation 1, and mean errors and 95% CI were calculated. In addition, the R² of the measurements from the reference instruments and the corrected sensor measurements were calculated.

RESULTS

Summary of Measurements

In the training dataset, 10.8% of hourly averaged measurements made by the Aeroqual O₃ sensor were < its LOD of 1 ppb. In the testing dataset, 42.9% of hourly averaged measurements made by the Aeroqual NO₂ sensor were < its LOD of 5 ppb, and 34.1% of hourly averaged measurements from the Aeroqual O₃ sensor were < its LOD of 1 ppb. Other sensor measurements were within their LOD. Hourly averaged data < LOD were removed.

Tables 1 and 2 present the summary statistics of hourly averaged NO₂, O₃, O₃+NO₂, T, and RH from the reference instruments and sensors in the training and testing datasets. In the training dataset (i.e., the dataset used to develop the regression models), ambient T ranged from 21.9 – 34.5°C, and ambient RH ranged from 39 – 90%. In the testing dataset (i.e., the dataset used to test the developed regression models), ambient T ranged from 7.4 – 28.4°C, and ambient RH ranged from 35 – 87%. The T and RH values were within the operating conditions specified by the manufacturers. The mean NO₂ as measured by NO_{2mon} was 4.6 ppb (range = 0.4 – 35 ppb) in the training dataset and 9.4 ppb (range = 1 – 37 ppb) in the testing dataset. Furthermore, the mean O₃ concentration as measured by O_{3mon} was 38.8 ppb (range = 1 – 65 ppb) in the training dataset and 35.7 ppb (range = 1 – 61 ppb) in the testing dataset.

Accuracy and Correlation During the First Co-Location Period

Table 3 and Figure S2 present the results obtained from the first co-location of the sensors with reference instruments at the monitoring station. During the first co-locating period, when the sensors were deployed to obtain data for the corrective models, there was a consistent positive bias in sensor measurements. Results of raw measurements from the Aeroqual O₃ sensor when compared to O_{3mon} had a mean error of 30% (95% CI = 13, 78) [Table 3, Row A, and Figure S2 A and B]. The variation in raw measurements of the Aeroqual O₃ sensor explained 71% of the variation in O_{3mon}.

Raw measurements from the Aeroqual NO₂ sensor when compared to NO_{2mon} had a mean error of 4264% (95% CI = 3160, 5662) [Table 3, Row B, and Figure S2 C and D]. The variation in raw measurements of the Aeroqual NO₂ sensor explained only 3% of the variation in NO_{2mon}. Raw measurements from the Cairclip O₃/NO₂ sensor when compared to (O₃+NO₂)_{mon} had a mean error of 65% (95% CI = 58, 76) [Table 3, Row C, and Figure S2 E and F]. Furthermore, the variation of raw measurements of the Cairclip O₃/NO₂ sensor explained 32% of the variation in (O₃+NO₂)_{mon}.

It appeared that higher RH and lower T were associated with lower measurements of O₃ and NO₂ (Figure S2). However, results of the variable selection for the final model showed that the effect of T on NO_{2mon} or O_{3mon} was not significant (results not shown). Consequently, T was not included in the final models (Table 4).

The final model for correcting the measured cofactors of the Aeroqual O₃ sensor performance included the following independent variables: The Aeroqual O₃ sensor squared,

$\text{NO}_{2\text{mon}}$ and RH^2 (relative humidity²) (Table 4A). In the model developed, measurements of Aeroqual O_3 sensor and RH had a non-linear relationship with $\text{O}_{3\text{mon}}$ (Table 4A).

The final model for correcting the measured cofactors of the Aeroqual NO_2 sensor performance included the measurements of the Aeroqual NO_2 sensor, $\text{O}_{3\text{mon}}$, and RH as independent variables (Table 4B). One ppb increase in NO_2 measured by the Aeroqual NO_2 sensor was associated with 0.1 ppb increase in $\text{NO}_{2\text{mon}}$ (Table 4B).

Lastly, the final model for correcting the measured cofactors of the Cairclip O_3/NO_2 sensor performance included the measurements of the Cairclip O_3/NO_2 sensor and RH as independent variables (Table 4C). One ppb increase in O_3+NO_2 measured by the Cairclip O_3/NO_2 sensor was associated with 0.1 ppb increase in $(\text{O}_3+\text{NO}_2)_{\text{mon}}$ (Table 4C).

Accuracy and Correlation in the Testing Dataset and Obtaining Corrected Readings from the Sensors

Table 5 and Figures S3 – S5 presents the results obtained from co-locating the sensors with reference instruments at the monitoring station in the testing dataset. Prior to the correction of the Aeroqual O_3 sensor, raw measurements of the Aeroqual O_3 sensor when assessed for accuracy against $\text{O}_{3\text{mon}}$, had a mean error of -9% (95% CI = $-18, 3$) in the testing dataset [Table 5, Row A, and Figure S3 A and B]. After testing the corrective model that controlled for measured cofactors of the Aeroqual O_3 sensor performance (NO_2 and RH) and the inherent error in the measurements, the corrected measurements of the Aeroqual O_3 sensor when assessed for accuracy against $\text{O}_{3\text{mon}}$, had a mean error of -1% (95% CI = $-14, 9$) [Table 5, Row A, and Figure S3 C and D]. Prior to the correction, the variation of raw measurements from the Aeroqual O_3 sensor explained 74% of the variation of $\text{O}_{3\text{mon}}$ in the testing dataset. After correction, the variation of the corrected Aeroqual O_3 sensor measurements explained 80% of the variation of $\text{O}_{3\text{mon}}$.

The raw measurements of the Aeroqual NO_2 sensor when assessed for accuracy against $\text{NO}_{2\text{mon}}$, had a mean error of 150% (95% CI = 98, 212) in the testing dataset [Table 5, Row B, and Figure S4 A and B]. After controlling for measured cofactors (O_3 and RH) and the inherent error in the measurements, the corrected measurements of the Aeroqual NO_2 sensor when assessed for accuracy against $\text{NO}_{2\text{mon}}$, had a mean error of -30% (95% CI = $-48, -10$) [Table 5, Row B, and Figure S4 C and D]. The variation of raw measurements from the Aeroqual NO_2 sensor explained only 8% of the variation of $\text{NO}_{2\text{mon}}$. After correction, the variation of the corrected Aeroqual NO_2 sensor measurements explained 24% of the variation of $\text{NO}_{2\text{mon}}$.

The raw measurements of the Cairclip O_3/NO_2 sensor when assessed for accuracy against $(\text{O}_3+\text{NO}_2)_{\text{mon}}$, had a mean error of 24% (95% CI = 20, 29) in the testing dataset [Table 5, Row C, and Figure S5 A and B]. After controlling for RH and the inherent error in the sensor measurements, the corrected measurements of the Cairclip O_3/NO_2 sensor when assessed for accuracy against $(\text{O}_3+\text{NO}_2)_{\text{mon}}$, had a mean error of 14% (95% CI = 9, 18) [Table 5, Row D, and Figure S5 C and D]. The variation of raw measurements from the Cairclip O_3/NO_2 sensor explained 63% of the variation of $(\text{O}_3+\text{NO}_2)_{\text{mon}}$. After correction,

the variation of the corrected Cairclip O₃/NO₂ sensor measurements explained 65% of the variation of (O₃+NO₂)_{mon}.

DISCUSSION

Results from this study show that none of the measurements of the three tested sensors, except for the Aeroqual O₃ sensor in the testing dataset, had a 95% CI of mean error within $\pm 25\%$. Thus, for the raw sensor measurements, only the Aeroqual O₃ sensor measurements in the testing dataset were within the NIOSH accuracy criterion (95% CI of the mean error = $-18 - 3\%$). However, in the training dataset, the Aeroqual O₃ sensor had a 95% CI of the mean error ranging from $13 - 78\%$ (outside $\pm 25\%$). Because 25% is included in the observed $13 - 78\%$ range, the data show that the accuracy of the Aeroqual O₃ sensor is varied, and may be accurate only under specific conditions. After controlling for NO₂, RH, and the inherent error in the Aeroqual O₃ sensor measurements, corrected measurements of the Aeroqual O₃ sensor improved in accuracy (mean error = -1%). This finding suggests that utilizing a calibration model that controls the effect of RH and NO₂ on the performance of the Aeroqual O₃ sensor may be required prior its use for occupational exposure assessment. The inaccurate results of the raw sensor measurements may not be due to sensor drift, as the sensors were within the recommended annual factory calibration cycle given by the manufacturers.

Comparison of measurements of the Cairclip O₃/NO₂ sensor with ambient O₃+NO₂ concentrations after controlling for RH and the inherent error in the sensor measurements showed accurate results (mean error = 14% [95% CI = 9–18%]). Conversely, measurements of the Aeroqual NO₂ sensor were not accurate, even after controlling for cofactors of the Aeroqual NO₂ sensor performance and the inherent error in measurements. The data suggest that Aeroqual NO₂ sensor may not be an accurate instrument for monitoring outdoor NO₂ exposures.

Because RH was a significant cofactor of the sensors performance and T was as not, the impact of varying levels of RH on the sensors performance is more important than fluctuating T. This indicates that RH needs to be measured in parallel with the Aeroqual O₃ and Cairclip O₃/NO₂ sensors when they are used for exposure assessment. In addition, simultaneous measurements of NO₂ may be needed when the Aeroqual O₃ sensor is deployed for measurements. As controlling for the effect of varying levels of RH was sufficient enough to make the Cairclip O₃/NO₂ sensor measurements within $\pm 25\%$ of O₃+NO₂, it is possible that the Cairclip O₃/NO₂ sensor measures accurately in environments that have low RH variation.

Our results show unequal variance of the tested sensors when compared to the corresponding reference instruments. In the training dataset, O_{3mon} and the Aeroqual O₃ sensor had similar standard deviations [SD] (~ 15 ppb). However, this was not the case in the testing dataset (O_{3mon} SD ~ 15 ppb and Aeroqual O₃ SD ~ 18 ppb). Furthermore, the SD of NO_{2mon} and the Aeroqual NO₂ sensor were markedly different in the training dataset, but approximately equal (8 ppb) in the testing dataset. Similarly, the Cairclip O₃/NO₂ sensor and the reference (O₃+NO₂)_{mon} instruments had markedly different SD in the training dataset, but similar SD

in the testing dataset. The data clearly show that the performance of the sensors is influenced by environmental conditions such as RH, which was controlled in the corrective models. However, it must be noted that other unmeasured cofactors of the sensors performance may exist.

Zimmerman et al. discovered that the response of one electrochemical NO₂ sensor (Alphasense ID: NO₂-B43F) was influenced by T, RH, CO, CO₂, SO₂, and O₃.⁽³⁴⁾ In that same study, RH had the greatest impact on the response of the NO₂ sensor in comparison to T, CO, CO₂, SO₂, and O₃.⁽³⁴⁾ In the current study, the linear regression model developed for the Aeroqual NO₂ sensor showed that the effect of O₃ on the performance of the Aeroqual NO₂ sensor was the greatest (i.e., greater magnitude of regression coefficient for O₃ in comparison to RH). This difference between the current study and that of Zimmerman et al.⁽³⁴⁾ could be due to the use of different NO₂ sensors (i.e., Alphasense ID: NO₂-B43F versus the Aeroqual NO₂ sensor) that have different levels of sensitivity to cofactors (RH and O₃, for example). RH values in the study by Zimmerman et al. was not reported, however, O₃ concentration was approximately 0 – 42 ppb,⁽³⁴⁾ and 1 – 65 ppb in the current study.

Other researchers have also reported the effects of cofactors of electrochemical sensor performance in the field.^(32, 33) Although controlling the influence of cofactors of sensor performance has been shown to improve the correlation of the corrected sensor measurements to reference/equivalent instruments, previous studies did not include comparison to the NIOSH $\pm 25\%$ accuracy criterion^(32–34). Therefore, it is unclear whether the tested sensors in the quoted studies^(32–34) are suitable for the use of occupational exposure assessment. Furthermore, the practical benefit of the sensors for exposure assessment may not be economical, given that measurements from other monitoring devices such as reference instruments are required to obtain accurate measurements using corrective models. To obtain accurate measurements from so-called low-cost sensors, modifications in hardware and operating technique may be required.

Limitations

We used relatively short co-locating periods for the instruments. Nevertheless, hourly fluctuations of T and RH during the three co-locating periods were representative of different weather conditions (for the seasons that we measured). As a result, we were able to assess the accuracy of the sensors in the field during different representative outdoor conditions. Restricting our analysis to measurements \geq LOD of the tested sensors can potentially cause a selection bias where data analyzed are not representative of environmental conditions associated with very low (i.e., $<$ LOD) sensor measurements. However, this method was employed in the current study in order to attain an unequivocal assessment of the performance of the tested sensors in the field. The operators of the monitoring station were only able to provide hourly averages of NO₂ and O₃, and for this reason the removal of measurements $<$ LOD were made after calculating hourly averages. The removal of raw measurements (i.e., data logged per minute) $<$ LOD is a more conservative approach to ensure the validity of data. Finally, the only potentially interfering cofactors that we measured were RH and T; there may be additional contributions from other atmospheric pollutants.

CONCLUSIONS

Raw measurements from the Aeroqual O₃ & NO₂ and Cairclip O₃/NO₂ sensors may be unsuitable for exposure assessment of outdoor workers such as traffic controllers, toll booth workers, and commercial drivers. Utilizing models to correct for cofactors of sensor performance are required to ensure accurate occupational exposure assessment. The practical benefit of these so-called low-cost sensors may not be economical, given that measurements from other monitoring devices such as reference/equivalent instruments are required to obtain accurate measurements from corrective models.

RECOMMENDATIONS

We recommend co-locating NO₂ and O₃ sensors with validated reference instruments to investigate the accuracy of sensor measurements before using them for occupational exposure assessment.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Table 1.

Summary of measurements in the training dataset.

	T (°C)	RH (%)	O_{3mon} (ppb)	Aeroqual O₃ (ppb)	NO_{2mon} (ppb)	Aeroqual NO₂ (ppb)	(O₃+NO₂)_{mon} (ppb)	Cairclip O₃/NO₂ (ppb)
Minimum	21.9	39.0	1.0	1.7	0.4	7.6	21.0	31.6
Maximum	34.5	90.0	65.0	69.7	35.0	92.9	69.0	187.6
Mean	27.4	69.8	38.8	43.1	4.6	30.5	43.4	70.0
SD	4.0	13.5	14.5	14.9	6.0	11.3	11.4	24.5
n	120	120	120	107	120	120	120	120

Measurements are based on hourly averages, training dataset = data used to develop the regression models. O_{3mon} = O₃ measured by the EPA reference-equivalent instrument with a photometric O₃ analyzer, Aeroqual O₃ = O₃ measured by the Aeroqual GSS O₃ sensor, NO_{2mon} = NO₂ measured by the EPA reference instrument with a NO_x chemiluminescence analyzer, Aeroqual NO₂ = NO₂ measured by the Aeroqual GSE NO₂ sensor, (O₃+NO₂)_{mon} = the sum of NO₂ and O₃ measured by the EPA reference instruments, Cairclip O₃/NO₂ = O₃+NO₂ measured by the Cairclip O₃/NO₂ sensor, n = number of observations.

Table 2.

Summary of measurements in the testing dataset.

	T (°C)	RH (%)	O_{3mon} (ppb)	Aeroqual O₃ (ppb)	NO_{2mon} (ppb)	Aeroqual NO₂ (ppb)	(O₃+NO₂)_{mon} (ppb)	Cairclip O₃/NO₂ (ppb)
Minimum	7.4	35.0	1.0	1.4	1.0	5.2	21.0	21.9
Maximum	28.4	87.0	61.0	63.1	37.0	31.2	68.0	86.1
Mean	17.5	64.5	35.7	33.1	9.4	15.0	45.1	47.3
SD	5.8	15.2	14.9	17.6	7.7	7.6	12.0	12.8
n	91	91	91	60	91	52	91	91

Measurements are based on hourly averages, testing dataset = data used to test the developed regression models. See footnotes of Table 1.

Table 3.

Accuracy and correlation of raw data from the NO₂ and O₃ sensors obtained from the training data set (data used to develop the regression models).

s/n	Comparison	n	R ²	Mean error	95% CI of mean error	Within NIOSH accuracy criterion
A.	Aeroqual O ₃ vs. O _{3mon}	107	0.71	30%	13 – 78%	No
B.	Aeroqual NO ₂ vs. NO _{2mon}	120	0.03	4264%	3160 – 5662%	No
C.	Cairclip O ₃ /NO ₂ vs. (O ₃ +NO ₂) _{mon}	120	0.32	65%	58 – 76%	No

n = number of observations, R² = coefficient of determination based on linear regression, mean error = percent difference between the measurements made by the sensors and measurements from the reference instruments (equation 1), 95% CI of mean error was obtained from BCa bootstrap confidence intervals, NIOSH accuracy criterion refers to 95% CI estimated mean error of ± 25%, Aeroqual O₃ = O₃ measured by the Aeroqual GSS O₃ sensor, O_{3mon} = O₃ measured by the EPA reference-equivalent instrument with a photometric O₃ analyzer, Aeroqual NO₂ = NO₂ measured by the Aeroqual GSE NO₂ sensor, NO_{2mon} = NO₂ measured by the EPA reference instrument with a NO_x chemiluminescence analyzer, Cairclip O₃/NO₂ = O₃+NO₂ measured by the Cairclip O₃/NO₂ Sensor, (O₃+NO₂)_{mon} = the sum of NO₂ and O₃ measured by the EPA reference instruments.

Table 4.

Results of the models for correcting the effect of measured cofactors of sensor performance.

A. Aeroqual O ₃ Sensor ^A		B. Aeroqual NO ₂ Sensor ^B		C. Cairclip O ₃ /NO ₂ Sensor ^C	
Parameters	Regression estimate	Parameters	Regression estimate	Parameters	Regression estimate
Intercept	46.974	Intercept	37.149	Intercept	70.474
(Aeroqual O ₃) ²	0.005	Aeroqual NO ₂	0.147	Cairclip O ₃ /NO ₂	0.139
NO ₂ _{mon}	-0.936	O ₃ _{mon}	-0.518		
(RH) ²	-0.003	RH	-0.245	RH	-0.534

^A response variable = O₃_{mon}. R² = 0.87; ^A result of equation S1 (n = 107)

^B response variable = NO₂_{mon}. R² = 0.66; ^B result of equation S2 (n = 120)

^C response variable = (O₃+NO₂)_{mon}. R² = 0.69; ^C result of equation S3 (n = 120)

See Table 3 for other footnotes.

Accuracy and correlation of corrected NO₂ and O₃ sensor readings obtained from the testing data set (data used to test the regression models).

Table 5.

s/n	Comparison	n	Action	R ²	Mean error	95% CI of mean error	Within NIOSH accuracy criterion
A.	Aeroqual O ₃ vs O _{3mon}	60	Raw	0.74	-9%	-18 – 3%	Yes
			Testing model in Table 4A	0.80	-1%	-14 – 9%	Yes
B.	Aeroqual NO ₂ vs. NO _{2mon}	52	Raw	0.08	150%	98 – 212%	No
			Testing model in Table 4B	0.24	-30%	-48 – -10%	No
C.	Cairclip O ₃ /NO ₂ vs. (O ₃ +NO ₂) _{mon}	91	Raw	0.63	24%	20 – 29%	No
			Testing model in Table 4C	0.65	14%	9 – 18%	Yes

See Table 3 for footnotes.