NIOSH Skin Notation Profiles Aniline



DEPARTMENT OF HEALTH AND HUMAN SERVICES Centers for Disease Control and Prevention National Institute for Occupational Safety and Health



This page intentionally left blank.

NIOSH Skin Notation (SK) Profile

Aniline [CAS No. 62-53-3]

Naomi L. Hudson and G. Scott Dotson

Department of Health and Human Services Centers for Disease Control and Prevention National Institute for Occupational Safety and Health This document is in the public domain and may be freely copied or reprinted.

Disclaimer

Mention of any company or product does not constitute endorsement by the National Institute for Occupational Safety and Health (NIOSH). In addition, citations to websites external to NIOSH do not constitute NIOSH endorsement of the sponsoring organizations or their programs or products. Furthermore, NIOSH is not responsible for the content of these websites. All Web addresses referenced in this document were accessible as of the publication date.

Ordering Information

To receive this document or information about other occupational safety and health topics, contact NIOSH:

Telephone: 1-800-CDC-INFO (1-800-232-4636) TTY: 1-888-232-6348 CDC-INFO: www.cdc.gov/info

or visit the NIOSH website at www.cdc.gov/niosh.

For a monthly update on news at NIOSH, subscribe to *NIOSH eNews* by visiting **www.cdc.gov/niosh/eNews.**

Suggested Citation

NIOSH [2015]. NIOSH skin notation profile: Aniline. By Hudson NL, Dotson GS. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2015-192

DHHS (NIOSH) Publication No. 2015-192

July 2015

SAFER • HEALTHIER • PEOPLETM

Foreword

As the largest organ of the body, the skin performs multiple critical functions, such as serving as the primary barrier to the external environment. For this reason, the skin is often exposed to potentially hazardous agents, including chemicals, which may contribute to the onset of a spectrum of adverse health effects ranging from localized damage (e.g., irritant contact dermatitis and corrosion) to induction of immune-mediated responses (e.g., allergic contact dermatitis and pulmonary responses), or systemic toxicity (e.g., neurotoxicity and hepatoxicity). Understanding the hazards related to skin contact with chemicals is a critical component of modern occupational safety and health programs.

In 2009, the National Institute for Occupational Safety and Health (NIOSH) published *Cur*rent Intelligence Bulletin (CIB) 61: A Strategy for Assigning New NIOSH Skin Notations [NIOSH 2009-147]. This document provides the scientific rationale and framework for the assignment of multiple hazard-specific skin notations (SK) that clearly distinguish between the systemic effects, direct (localized) effects, and immune-mediated responses caused by skin contact with chemicals. The key step within assignment of the hazard-specific SK is the determination of the hazard potential of the substance, or its potential for causing adverse health effects as a result of skin exposure. This determination entails a health hazard identification process that involves use of the following:

- · Scientific data on the physicochemical properties of a chemical
- Data on human exposures and health effects
- Empirical data from *in vivo* and *in vitro* laboratory testing
- Computational techniques, including predictive algorithms and mathematical models that describe a selected process (e.g., skin permeation) by means of analytical or numerical methods.

This *Skin Notation Profile* provides the SK assignments and supportive data for aniline. In particular, this document evaluates and summarizes the literature describing the hazard potential of the substance and its assessment according to the scientific rationale and framework outlined in CIB 61. In meeting this objective, this *Skin Notation Profile* intends to inform the audience—mostly occupational health practitioners, researchers, policy- and decision-makers, employers, and workers in potentially hazardous workplaces—so that improved risk-management practices may be developed to better protect workers from the risks of skin contact with the chemicals of interest.

> John Howard, M.D. Director, National Institute for Occupational Safety and Health Centers for Disease Control and Prevention

This page intentionally left blank.

Contents

Foreword	iii
Abbreviations	vi
Glossary	viii
Acknowledgments	ix
1 Introduction	1
1.1 General Substance Information	1
1.2 Purpose	1
1.3 Overview of SK Assignment	1
2 Systemic Toxicity from Skin Exposure (SK: SYS)	2
3 Direct Effects on Skin (SK:DIR)	4
4 Immune-mediated Responses (SK: SEN)	4
5 Summary	5
Appendix: Calculation of the SI Ratio	8
Overview	8
Calculation	9
Appendix References	9

Abbreviations

ACGIH	American Conference of Governmental Industrial Hygienists
ATSDR	Agency for Toxic Substances and Disease Registry
CIB	Current Intelligence Bulletin
cm ²	square centimeter(s)
cm/hr	centimeter(s) per hour
cm/s	centimeter(s) per second
DEREK	Deductive Estimation of Risk from Existing Knowledge
DIR	skin notation indicating the potential for direct effects to the skin following
DIK	contact with a chemical
EC	European Commission
g	gram(s)
g/L	gram(s)/liter
GHS	Globally Harmonized System for Classification and Labelling of Chemicals
GPMT	guinea pig maximization test
hr	hour(s)
IARC	International Agency for Research on Cancer
IPCS	International Program for Chemical Safety
(IRR)	subnotation of SK: DIR indicating the potential for a chemical to be a skin irritant following exposure to the skin
k _{aq}	coefficient in the watery epidermal layer
k _p	skin permeation coefficient
k_{pol}	coefficient in the protein fraction of the stratum corneum
k_{psc}	permeation coefficient in the lipid fraction of the stratum corneum
LD_{50}	dose resulting in 50% mortality in the exposed population
LD _{Lo}	dermal lethal dose
LLNA	local lymph node assay
LOAEL	lowest-observed-adverse-effect level
$\log K_{_{OW}}$	base-10 logarithm of a substance's octanol–water partition
M	molarity
m ³	cubic meter(s)
mg	milligram(s)
mg/cm²/hr	milligram(s) per square centimeter per hour
mg/kg	milligram(s) per kilogram body weight
mg/m ³	milligram(s) per cubic meter
mL	milliliter(s)
mL/kg	milliliter(s) per kilogram body weight
MW	molecular weight
NIOSH	National Institute for Occupational Safety and Health
NOAEL	no-observed-adverse-effect level

NTP	National Toxicology Program
OEL	occupational exposure limit
OSHA	Occupational Safety and Health Administration
ppm	parts per million
REL	recommended exposure limit
RF	retention factor
SEN	skin notation indicating the potential for immune-mediated reactions following exposure of the skin
SI ratio	ratio of skin dose to inhalation dose
SK	skin notation
S_{W}	solubility in water
SYS	skin notation indicating the potential for systemic toxicity following exposure of the skin
US EPA	United States Environmental Protection Agency
μg	microgram(s)
µg/cm ²	microgram(s) per square centimeter
µg/cm²/hr	microgram(s) per square centimeter per hour
μL	microliter(s)
μmol	micromole(s)

Glossary

Absorption—The transport of a chemical from the outer surface of the skin into both the skin and systemic circulation (including penetration, permeation, and resorption).

Acute exposure—Contact with a chemical that occurs once or for only a short period of time.

Cancer—Any one of a group of diseases that occurs when cells in the body become abnormal and grow or multiply out of control.

Contaminant—A chemical that is (1) unintentionally present within a neat substance or mixture at a concentration less than 1.0% or (2) recognized as a potential carcinogen and present within a neat substance or mixture at a concentration less than 0.1%.

Cutaneous (or percutaneous)—Referring to the skin (or through the skin).

Dermal—Referring to the skin.

Dermal contact—Contact with (touching) the skin.

Direct effects—Localized, non-immune-mediated adverse health effects on the skin, including corrosion, primary irritation, changes in skin pigmentation, and reduction/ disruption of the skin barrier integrity, occurring at or near the point of contact with chemicals.

Immune-mediated responses—Responses mediated by the immune system, including allergic responses.

Sensitization—A specific immune-mediated response that develops following exposure to a chemical, which, upon re-exposure, can lead to allergic contact dermatitis (ACD) or other immune-mediated diseases such as asthma, depending on the site and route of re-exposure.

Substance—A chemical.

Systemic effects—Systemic toxicity associated with skin absorption of chemicals after exposure of the skin.

Acknowledgments

This document was developed by the NIOSH Education and Information Division (Paul Schulte, Ph.D., Director). G. Scott Dotson, Ph.D., was the project officer for this document, assisted in great part by Naomi Hudson, Dr.P.H., MPH; Stacey Anderson, Ph.D.; and Matt Dahm, M.Sc. The basis for this document was a report (*Toxicology Excellence for Risk Assessment [TERA]*) contracted by NIOSH and prepared by Bernard Gadagbui, Ph.D. and Andrew Maier, Ph.D.

For their contribution to the technical content and review of this document, special acknowledgment is given to the following NIOSH personnel:

Denver Field Office

Eric Esswein, M.Sc.

Division of Applied Research and Technology

Clayton B'Hymer, Ph.D. John Snawder, Ph.D. Mark Toraason, Ph.D.

Division of Respiratory Disease Studies

Gregory A. Day, Ph.D. Aleksander Stefaniak, Ph.D.

Division of Surveillance, Hazard Evaluations, and Field Studies

Todd Niemeier, M.Sc. Aaron Sussell, Ph.D. Loren Tapp, M.D.

Education and Information Division

Devin Baker, M.Ed. Charles L. Geraci, Ph.D. Thomas J. Lentz, Ph.D. Richard Niemeier, Ph.D. Ralph Zumwalde, M.Sc.

Health Effects Laboratory Division

H. Fredrick Frasch, Ph.D. Vic Johnson, Ph.D. Michael Luster, Ph.D. Anna Shvedova, Ph.D. Paul Siegel, Ph.D. Berran Yucesoy, Ph.D.

National Personal Protection Technology Laboratory

Heinz Ahlers, M.Sc. Angie Shephard For their contribution to the technical content and review of this document, special acknowledgment is given to the following CDC personnel:

Office of Surveillance, Epidemiology and Laboratory Services/Epidemiology and Analysis Program Office

Barbara Landreth, M.A.

In addition, special appreciation is expressed to the following individuals for serving as independent, external reviewers and providing comments that contributed to the development or improvement of this document:

Phil Bigelow, B.Sc., M.H.Sc., Ph.D., University of Waterloo, School of Public Health and Health Systems, Waterloo, ON, Canada

Glenn Sipes, Ph.D., University of Arizona, College of Medicine, Tucson, AZ

G. Frank Gerberick, Ph.D., The Procter and Gamble Company, Cincinnati, OH

Dori Germolec, Ph.D., National Toxicology Program, National Institute for Environmental Health Sciences, Research Triangle, NC

Ben Hayes, M.D., Ph.D., Division of Dermatology, Vanderbilt School of Medicine, Nashville, TN

Jennifer Sahmel, M.Sc., CIH, ChemRisk, Boulder, CO

James Taylor, M.D., Industrial Dermatology, The Cleveland Clinic, Cleveland, OH

1 Introduction

1.1 General Substance Information

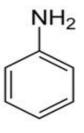
Chemical: Aniline

CAS No: 62-53-3

Molecular weight (MW): 93.1

Molecular formula: C₆H₅NH₂

Structural formula:



1.2 Purpose

This skin notation profile presents (1) a brief summary of epidemiological and toxicological data associated with skin contact with aniline and (2) the rationale behind the hazardspecific skin notation (SK) assignment for aniline. The SK assignment is based on the scientific rationale and logic outlined in the Current Intelligence Bulletin (CIB) 61: A Strategy for Assigning New NIOSH Skin Notations [NIOSH 2009]. The summarized information and health hazard assessment are limited to an evaluation of the potential health effects of dermal exposure to aniline. A literature search was conducted through August 2014 to identify information on aniline, including but not limited to data relating to its toxicokinetics,

Synonyms:

Aminobenzene, Aniline oil, Benzenamine, Phenylamine

Uses:

Aniline is primarily used in the manufacturing of dyestuffs, as a chemical intermediate, and as a rubber accelerator [Weisburger and Hudson 2001]. In addition, an estimated 900 million pounds (~409 million kilograms) of aniline were produced in 1992 [Weisburger and Hudson 2001].

acute toxicity, repeated-dose systemic toxicity, carcinogenicity, biological system/function– specific effects (including reproductive and developmental effects and immunotoxicity), irritation, and sensitization. Information was considered from studies of humans, animals, or appropriate modeling systems that are relevant to assessing the effects of dermal exposure to aniline.

1.3 Overview of SK Assignment

Aniline is potentially capable of causing numerous adverse health effects following skin contact. A critical review of available data has resulted in the following SK assignment for aniline: **SK: SYS-SEN**. Table 1 provides an

Skin notation	notation Critical effect Available data	
SK: SYS	Methemoglobinemia Limited human and sufficient anim	
SK: SEN	: SEN Skin allergy Limited human and sufficient animal d	

overview of the critical effects and data used to develop the SK assignment for aniline.

2 Systemic Toxicity from Skin Exposure (SK: SYS)

Toxicokinetic studies following dermal exposure to aniline in humans and animals were identified. Dutkiewicz [1961] conducted 20 experiments to determine the potential of dermal absorption in humans exposed to aniline vapor for up to 5 hours. Dutkiewicz [1961] reported an average dermal absorption of 25.6-33 micrograms per square centimeter per hour (µg/cm²/hr) and 15.0 µg/cm²/hr with use of work clothing (i.e. overalls). In an industrial setting, Dutkeiwicz [1961] reported that 26.5% and 50% of the dose of aniline were dermally absorbed from vapor and liquid exposures, respectively. Other factors reported to affect percutaneous absorption of aniline included temperature, moisture, humidity, exposure duration and concentration of aniline applied [Piotrowski 1957; Dutkiewicz 1961]. Baranowska-Dutkiewicz [1982] found that absorption rates of aniline increased with evaluated concentrations and decreased with longer exposure times when 10 volunteers were exposed to solutions of 1-2% aniline for 30 or 60 minutes. Baranowska-Dutkiewicz [1982] also reported dermal absorption rates of 3.0 milligrams per square centimeter per hour (mg/cm²/hr) and 2.5 mg/cm²/hr from liquid aniline and aniline with 3% water, and 0.20- $1.22 \text{ mg/cm}^2/\text{hr}$ from aqueous solution (1–2%) in water), following 30 or 60 minute dermal applications of dilute aqueous solutions of aniline to human skin in vivo. Piotrowski [1957] reported dermal absorption rates of 0.18 to 0.72 mg/cm²/hr in 15 tests conducted on 11 human subjects exposed to aniline on the forearm under gauze for 5 hours and 3.8 mg/ cm²/hr when the gauze was moistened with water and the subjects exposed for 1 hour. In mice, Susten et al. [1990] calculated a dermal absorption of 4.7% of the applied dose and an absorption rate of 2.3 mg/cm²/min.

In vitro studies using human and animal skin were also identified. Barry et al. [1985] reported that permeation of aniline in vitro through dermatomed human abdominal cadaver skin was 1870 µg/cm²/hr for pure liquid, 760 µg/ cm^2/hr for saturated aqueous solution, 260 µg/ cm²/hr for the vapor of the pure liquid, and 250 µg/cm²/hr for vapor of saturated aqueous solution. Korinth et al. [2012] reported the flux of aniline was 725.2 μ g/cm²/hr using Franz diffusion cells in close agreement with OECD guideline 428. The potential of aniline to pose a skin absorption hazard was also evaluated, with use of a predictive algorithm for estimating and evaluating the health hazards of dermal exposure to substances [NIOSH 2009]. The evaluation method compares an estimated dose accumulated in the body from skin absorption and an estimated dose from respiratory absorption associated with a reference occupational exposure limit. On the basis of this algorithm, a ratio of the skin dose to the inhalation dose (SI ratio) of 2.28 was calculated for aniline. An SI ratio of ≥0.1 indicates that skin absorption may significantly contribute to the overall body burden of a substance [NIOSH 2009]; therefore, aniline is considered to be absorbed through the skin following dermal exposure. Additional information on the SI ratio and the variables used in its calculation are included in the appendix.

No dermal lethal doses (LD_{Lo}'s) of aniline in humans have been identified. The dermal LD_{50} value (the dose resulting in 50% mortality in the exposed animals) was 0.82 mL/kg [corresponding to 838 mg/kg, based on density of 1.0217 g/mL] when applied to abraded rabbit skin and 2.15 ml/kg [corresponding to 2200 mg/kg] and 1.29 ml/kg [corresponding to 1318 mg/kg] when applied to abraded and intact guinea pig skin, respectively [Roudabush et al. 1965]. Dow Chemical Company [1940] found dermal application of 2400 mg/ kg to produce no lethality when administered to rabbits, while doses greater than 3000 mg/ kg produced 100% mortality. Because the reported acute dermal LD₅₀ values for guinea pigs and rabbits are generally lower than the critical dermal $\mathrm{LD}_{\mathrm{50}}$ value of 2000 mg/ kg body weight that identifies chemical substances with the potential for acute dermal toxicity [NIOSH 2009], aniline demonstrates acute toxicity following dermal exposure.

Numerous instances of accidental and occupational exposures to aniline have been reported. Two cases of accidental spraying of the skin with aniline reported symptoms including discoloring of the skin, weakness, headache, sinus tachycardia, and methemoglobin (MetHb) levels of up to 70% [Phillips et al. 1990; Cummings et al. 1994]. Liao et al. [2002] reported dermal exposure to aniline in a worker. The worker was treated with methylene blue; however, methemoglobin recurred, followed by severe Heinz body hemolytic anemia. As is typical of accidental or occupational exposures, the doses of aniline that elicited these effects have not been quantified. Lee et al. [2013] reported a worker dermally exposed to 200 cc of aniline that splattered on his face and upper body. The worker suffered from a burn around the exposed skin, and showed signs of cyanoderma on his entire body. The workers' MetHb level was 46.8%, and he was diagnosed with methemoglobinemia [Leeetal.2013].TheNationalResearchCouncil

[NRC 2000] reported aniline is absorbed through the skin and is a methemoglobinforming compound as seen in these case reports.

No repeat-dose, sub-chronic, or chronic studies of dermal exposure to aniline in humans or animals were identified. No standard toxicity or specialty studies evaluating biological system/function specific effects (including reproductive and developmental effects and immunotoxicity) following dermal exposure to aniline were identified in animals. No assessment of the carcinogenicity of aniline following dermal exposure has been identified. However, various organizations have evaluated the potential of aniline to be carcinogenic via other routes. Table 2 summarizes carcinogenic designations of multiple governmental and nongovernmental organizations for aniline.

Taken together, data from *in vivo* and *in vitro* toxicokinetic studies in humans [Piotrowski 1957; Dutkiweicz 1961; Baranowska-Dut-keiwicz 1982; Barry et al. 1985; Korinth et al. 2012] indicate that aniline is dermally absorbed. The acute dermal toxicity studies

Organization	Carcinogenic designation
NIOSH [2005]	Potential occupational carcinogen
NTP [2014]	No designation
USEPA [2014]	Group B2: probable human carcinogen
European Parliament [2008]	GHS Carcinogenicity Category 2: Suspected of causing cancer
IARC [2012]	Group 3: not classifiable as to carcinogenicity to humans
EC [2014] [†]	R40: limited evidence of a carcinogenic effect
ACGIH [2001]	Group A3: confirmed animal carcinogen with unknown relevance to humans

Table 2. Summary of the carcinogenic designations* for aniline by numerousgovernmental and nongovernmental organizations

ACGIH = American Conference of Governmental Industrial Hygienists; EC = European Commission, Joint Research, Institute for Health and Consumer Protection; GHS = Globally Harmonized System for Classification and Labelling of Chemicals; IARC = International Agency for Research on Cancer; NIOSH = National Institute for Occupational Safety and Health; NTP = National Toxicology Program; USEPA = United States Environmental Protection Agency.

*The listed cancer designations were based on data from nondermal (such as oral or inhalation) exposure rather than dermal exposure. *Date accessed. in rabbits and guinea pigs [Roudabush et al 1965]⁺, and human case reports [Phillips et al 1990; Cummings et al. 1994; Liao et al 2002; Lee et al. 2013] demonstrate that aniline is absorbed through the skin, and is systemically available and can cause methemoglobinemia. Therefore, on the basis of the data for this assessment, aniline is assigned the SK: SYS notation.

3 Direct Effects on Skin (SK:DIR)

No human or animal *in vivo* studies for corrosivity of aniline or *in vitro* tests for corrosivity using human or animal skin models or *in vitro* tests of skin integrity using cadaver skin were identified. No data were identified for humans upon which the skin irritation potential of aniline can be evaluated. In rabbits, Dow Chemical Company [1940] reported aniline produced slight skin irritation.

Data that evaluated potential of aniline to cause skin irritation in animals [Dow Chemical Company 1940] are insufficient to support an IRR notation for aniline. Therefore, on the basis of the data for this assessment, aniline is not assigned the SK: DIR (IRR) notation.

4 Immune-mediated Responses (SK: SEN)

A number of studies were identified that evaluated the potential of aniline to cause skin sensitization in humans and animals. A recent analysis of patch test results from a contact allergy surveillance network by Uter et al. [2007] observed positive allergic reactions to aniline in 25 of 119 patients. Twenty four of the 25 patients were also positive to *para*-phenylenediamine, *p*-aminoazobenzene and (in one case) another *para*-amino compound; however, the investigators' analysis of the individual clinical data could not reveal if the patients were exposed to aniline. Therefore, the investigators concluded that aniline was unlikely to be an independent sensitizer, but more likely to produce allergic reactions in individuals pre-sensitized to *para*-substituted amino compounds. In a human maximization test (HMT), Basketter et al. [1994] recorded positive responses in 28% of the subjects (unspecified number) induced with 20% aniline and later challenged with 10% aniline. In a database of inter-laboratory study results compiled by Haneke et al. [2001], aniline reportedly produced positive reactions in the HMT.

In animals, tests of sensitizing potential have produced equivocal results. Goodwin et al. [1981] evaluated the skin sensitization potential of aniline using three guinea pig sensitization procedures and observed positive allergic reactions in 1 of 10 animals (10%) in the guinea pig maximization test (GPMT), 5 of 10 animals (50%) in the single injection adjuvant test (SIAT), but none in the modified Draize test. Subsequently, Godwin et al. [1981] classified the potential of aniline to cause sensitization in the guinea pigs as weak in the GMPT and moderate in the SIAT, and a mild sensitizer based on their assessment of clinical data in humans. Basketter and Scholes [1992] classified aniline as an extreme sensitizer based on the 90% positive reactions in the GPMT, although the LLNA test was borderline. In a later study, Basketter et al. [2003] concluded that aniline was a weak sensitizer in the LLNA procedure. The Haneke et al. [2001] database reported aniline as a skin sensitizer in the guinea pig maximization test/Buehler assay; however, reviewing the results of the murine local lymph node assays (LLNA), negative responses were reported. The structure activity relationship model, DEREK for Windows, also predicted aniline to be positive regarding skin sensitization potential.

The results from the HMT [Basketter et al. 1994; Haneke et al. 2001], GPMTs [Goodwin et al. 1981; Basketter and Scholes 1992], and the weight of evidence from the murine LLNAs

^{*}References in bold text indicate studies that serve as the basis of the SK assignments.

[Basketter and Scholes 1992; Basketter et al. 2003], supported by the structureactivity relationship model prediction, demonstrate that aniline is a skin sensitizer. Therefore, on the basis of the data for this assessment, aniline is assigned the SK: SEN notation.

5 Summary

Although there is a paucity of data evaluating the systemic effects of chronic dermal exposure to aniline in humans and animals, in vivo and in vitro toxicokinetic studies in humans [Piotrowski 1957; Dutkiweicz 1961; Baranowska-Dutkeiwicz 1982; Barry et al. 1985; Korinth et al. 2012] indicate that aniline is dermally absorbed. The acute dermal toxicity studies in rabbits and guinea pigs [Roudabush et al 1965; Industrial Bio-Test Laboratories 1969], and human case reports [Phillips et al 1990; Cummings et al. 1994; Liao et al 2002; Lee et al. 2013] are sufficient to conclude that aniline is absorbed through the skin, is systemically available, and can produce methemoglobinemia. Data that evaluated potential of aniline to cause skin irritation in animals [Dow Chemical Company 1940] are insufficient to suggest that aniline produces mild skin irritation. There is sufficient data from human maximization tests [Basketter et al. 1994] guinea pig maximization tests [Goodwin et al. 1981; Basketter and Scholes 1992], murine LLNAs (based on the weight of evidence) from [Basketter and Scholes 1992; Basketter et al. 2003] to demonstrate that aniline is a skin sensitizer. Therefore, on the basis of these assessments, aniline is assigned a composite skin notation of SK: SYS-SEN.

Table 3 summarizes the skin hazard designations for aniline previously issued by NIOSH and other organizations. The equivalent dermal designations for aniline, according to the Global Harmonization System (GHS) of Classification and Labelling of Chemicals, are Acute Toxicity Category 3 (Hazard statement: Toxic in contact with the skin), and Skin Sensitization Category 1 (Hazard statement: May cause an allergic skin reaction) [European Parliament 2008]. In addition, aniline has been classified as a Mutagenicity Category 2 (Hazard Statement: Suspected of causing genetic defects) [European Parliament 2008].

References

Note: Asterisks (*) denote sources cited in text; daggers (†) denote additional resources.

*ACGIH [2001]. Aniline. In: Documentation of threshold limit values and biological exposure indices 7th ed., Vol. 1. Cincinnati, OH: American Conference of Governmental Industrial Hygienists.

Organization	Skin hazard designation
NIOSH [2005]	No designation
OSHA [2015]*	[skin]: potential for dermal absorption
ACGIH [2001]	[skin]: based on the increase in methemoglobin in blood in animals and skin absorption in humans
EC [2014]*	R24: toxic in contact with skin
	R43: may cause sensitization by skin contact

Table 3. Summary of previous skin hazard designations for aniline

ACGIH = American Conference of Governmental Industrial Hygienists; EC = European Commission, Joint Research, Institute for Health and Consumer Protection; NIOSH = National Institute for Occupational Safety and Health; OSHA = Occupational Safety and Health Administration.

*Date accessed.

- *Baranowska-Dutkiewicz B [1982]. Skin absorption of aniline from aqueous solutions in man. Toxicology Letters *10*(4):367–372.
- *Barry BW, Harrison SM, Dugard PH [1985]. Vapour and liquid diffusion of model penetrants through human skin; correlation with thermodynamic activity. Journal of Pharmacy and Pharmacology *37*:226–236.
- *Basketter DA, Scholes EW [1992]. Comparison of the local lymph node assay with the guineapig maximization test for the detection of a range of contact allergens. Food and Chemical Toxicology *30*:65–69.
- *Basketter DA, Scholes EW, Kimber I [1994]. The performance of the local lymph node assay with chemicals identified as contact allergens in the human maximization test. Food and Chemical Toxicology *32*(6):543–547.
- *Basketter DA, Smith Pease CK, Patlewicz GY [2003]. Contact allergy: the local lymph node assay for the prediction of hazard and risk. Clin Exp Dermatol 28:218–21.
- [†]Boeniger MF, Lushniak BD [2000]. Exposure and absorption of hazardous materials through the skin. Int J Occup Environ Health 6(2):148– 150.
- †Casciano AD [1952]. Acute Methemoglobinemia due to aniline—report of a case. Journal of the Medical Society of New Jersey 49(4):141– 142. [Abstract only]
- *Cummings MH, Day S, Norton S [1994]. Petrol octane booster and methaemoglobinaemia. J Int Med 235:279–280.
- *Dow Chemical Company [1940]. The toxicity of aniline, aniline hydrochloride, dimethyl aniline, and diethyl aniline. OTS 0558237
- *Dutkiewicz T [1961]. Absorption of aniline vapors in men. Proceedings of the 13th International Congress on Occupational Health. (Paper presented at New York, July 1960) pages 681–686. New York: Book Craftsmen Associates, Inc.
- *EC (European Commission) [ND]. Aniline. In: EINICS (European Inventory of Existing Commercial Chemical Substances), http://esis. jrc.ec.europa.eu/. Accessed: 8-18-14.
- *European Parliament, Council of the European Union [2008]. Regulation (EC) No 1272/2008 of the European Parliament and of the Council of 16 December 2008 on classification, labelling and packaging of substances and mixtures, amending and repealing Directives 67/548/ EEC and 1999/45/EC, and amending Regulation (EC) No 1907/2006. OJEU, Off J Eur

Union L353:1–1355, http://eur-lex.europa.eu/ LexUriServ/LexUriServ.do?uri=OJ:L:2008:35 3:0001:1355:EN:PDF. Accessed: 02-10-15.

- *Goodwin BFJ, Crevel RWR, Johnson AW [1981]. A comparison of three guinea-pig sensitization procedures for the detection of 19 reported human contact sensitizers. Contact Dermatitis 7:248–258.
- *Haneke KE, Tice RR, Carson BL, Margolin BH, Stokes WS [2001]. ICCVAM evaluation of the murine local lymph node assay. Data analyses completed by the National Toxicology Program Interagency Center for the Evaluation of Alternative Toxicological Methods. Regul Toxicol Pharmacol 34:274–86.
- 'Heim de Balsac F; Agasse Lafont E; Feil A [1928]. Investigation of chronic occupational anilinism in the imitation fur industry. Journal Medical Francais 17:377–380.
- *IARC (International Agency for Research on Cancer) [2012]. Agents reviewed by the IARC monographs. In: IARC monographs on the evaluation of carcinogenic risks to humans, http://monographs.iarc.fr/ENG/Monographs/ PDFs/index.php. Accessed: 02-10-15.
- *Korinth G, Schaller KH, Bader M, Bartsch R, Göen T, Roßbach B, Drexler H [2012]. Comparison of experimentally determined and mathematically predicted percutaneous penetration rates of chemicals. Arch Toxicol *86*(3): 423–430.
- *Lee CH, Kim SH, Kwon do H, Jang KH, Chung YH, Moon JD [2013]. Two cases of methemoglobinemia induced by the exposure to nitrobenzene and ainiline. Ann Occup Environ Med *25*(1):31.
- *Liao YP, Hung DZ, Yang DY [2002]. Hemolytic anemia after methylene blue therapy in aniline induced methemoglobinema. Vet Human Toxicol *44*(1):19–21.
- *NIOSH [2005]. Aniline. In: NIOSH pocket guide to chemical hazards. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2005-149, http://www.cdc.gov/ niosh/npg/. Accessed: 02-10-15.
- *NIOSH [2009]. Current intelligence bulletin 61: a strategy for assigning new NIOSH skin notations. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health,

DHHS (NIOSH) Publication No. 2009-147, http://www.cdc.gov/niosh/docs/2009-147/pdfs/2009-147.pdf. Accessed: 02-10-15.

- *NRC (National Research Council) [2000]. Acute exposure guideline levels for selected airborne chemicals: volume 1, http://www.nap. edu/catalog.php?record_id=10043. Accessed: 02-10-15.
- *NTP [2014]. Report on Carcinogens. Thirteenth Edition; U.S. Department of Health and Human Services, Public Health Service. National Toxicology Program, http://ntp.niehs.nih.gov/ pubhealth/roc/roc13/index.html. Accessed: 02-10-15.
- *OSHA [ND]. Aniline. In: OSHA/EPA occupational chemical database, http://www.osha.gov/ chemicaldata/chemResult.html?recNo=452. Accessed: 02-10-15.
- *Phillips DM, Gradisek R, Heiselman DE [1990]. Methemoglobinemia secondary to aniline exposure. Ann Emerg Med *19*:425–429.
- *Piotrowski J [1957]. Quantitative estimation of aniline absorption through the skin in man. Journal of Hygiene, Epidemiology, Microbiology and Immunology *1*:23–32.

- †Prout J [1973]. Allergic dermatitis due to aniline dye additive. Proceedings of the Royal Society of Medicine 66:261–262
- *Roudabush RL, Terhaar CJ, Fassett DW, Dziuba SP [1965]. Comparative acute effects of some chemicals on the skin of rabbits and guinea pigs. Toxicol Appl Pharmacol 7(4):559–65.
- *Susten AS, Niemeier RW, Simon SD [1990]. In vivo percutaneous absorption studies of volatile organic solvents in hairless mice. II. Toluene, Ethylbenzene, and Aniline. Journal of Applied Toxicology *10*(3):217–225.
- *US EPA (United States Environmental Protection Agency) [2014]. Aniline. Integrated Risk Information System. Washington, DC: United StatesEnvironmentalProtection Agency, http://www.epa.gov/iris/subst/0350.htm. Accessed: 02-10-15.
- *Uter W, Stropp G, Schnuch A, Lessmann H [2007]. Aniline—A 'historical' contact allergen? Current data from the IVDK and review of the literature. Ann Occup Hyg *51*:219–226.
- *Weisburger EK, Hudson VW [2001]. Aromatic nitro and amino compounds. In: Bingham E, Cohrssen B, Powell CH (Eds.), Patty's toxicology 5th ed., Vol. 4. New York: John Wiley & Sons, Inc. pp. 817–968.

Appendix: Calculation of the SI Ratio

This appendix presents an overview of the SI ratio and a summary of the calculation of the SI ratio for aniline. Although the SI ratio is considered in the determination of a substance's hazard potential following skin contact, it is intended only to serve as supportive data during the assignment of the NIOSH SK. An in-depth discussion on the rationale and calculation of the SI ratio can be found in Appendix B of the *Current Intelligence Bulletin (CIB) 61: A Strategy for Assigning New NIOSH Skin Notations [NIOSH 2009].*

Overview

The SI ratio is a predictive algorithm for estimating and evaluating the health hazards of skin exposure to substances. The algorithm is designed to evaluate the potential for a substance to penetrate the skin and induce systemic toxicity [NIOSH 2009]. The goals for incorporating this algorithm into the proposed strategy for assigning SYS notation are as follows:

- Provide an alternative method to evaluate substances for which no clinical reports or animal toxicity studies exist or for which empirical data are insufficient to determine systemic effects.
- 2. Use the algorithm evaluation results to determine whether a substance poses a skin absorption hazard and should be labeled with the SYS notation.

The algorithm evaluation includes three steps:

- Determining a skin permeation coefficient (k_p) for the substance of interest.
- 2. Estimating substance uptake by the skin and respiratory absorption routes.
- 3. Evaluating whether the substance poses a skin exposure hazard.

The algorithm is flexible in the data requirement and can operate entirely on the basis of the physicochemical properties of a substance and the relevant exposure parameters. Thus, the algorithm is independent of the need for biologic data. Alternatively, it can function with both the physicochemical properties and the experimentally determined permeation coefficient when such data are available and appropriate for use.

The first step in the evaluation is to determine the k_p for the substance to describe the transdermal penetration rate of the substance [NIOSH 2009]. The k_p , which represents the overall diffusion of the substance through the stratum corneum and into the blood capillaries of the dermis, is estimated from the compound's molecular weight (*MW*) and base-10 logarithm of its octanol-water partition coefficient (log K_{oW}). In this example, k_p is determined for a substance with use of Equation 1. A self-consistent set of units must be used, such as centimeters per hour (cm/hr), outlined in Table A1. Other model-based estimates of k_p may also be used [NIOSH 2009].

Equation 1: Calculation of Skin Permeation Coefficient (k,)

$$k_p = \frac{1}{\frac{1}{k_{psc} + k_{pol}} + \frac{1}{k_q}}$$

where k_{psc} is the permeation coefficient in the lipid fraction of the stratum corneum, k_{pol} is the coefficient in the protein fraction of the stratum corneum, and k_{aq} is the coefficient in the watery epidermal layer. These components are individually estimated by

$$\log K_{psc} = -1.326 + 0.6097 \times \log K_{ow} - 0.1786 \times MW^{0.5}$$
$$K_{pol} = 0.0001519 \times MW^{-0.5}$$

1

The second step is to calculate the biologic mass uptake of the substance from skin absorption (skin dose) and inhalation (inhalation

 $K_{ag} = 2.5 \times MW^{-0.5}$

dose) during the same period of exposure. The skin dose is calculated as a mathematical product of the k_p , the water solubility (S_w) of the substance, the exposed skin surface area, and the duration of exposure. Its units are milligrams (mg). Assume that the skin exposure continues for 8 hours to unprotected skin on the palms of both hands (a surface area of 360 square centimeters [cm²]).

Equation 2: Determination of Skin Dose

Skin dose = $k_p \times S_w \times$ Exposed skin surface area × Exposure time

$$= k_p(\text{cm/hour}) \times S_w(\text{mg/cm}^3) \times 360 \text{ cm}^2 \times 8 \text{ hours}$$

The inhalation dose (in mg) is derived on the basis of the occupational exposure limit (OEL) of the substance—if the OEL is developed to prevent the occurrence of systemic effects rather than sensory/irritant effects or direct effects on the respiratory tract. Assume a continuous exposure of 8 hours, an inhalation volume of 10 cubic meters (m³) inhaled air in 8 hours, and a factor of 75% for retention of the airborne substance in the lungs during respiration (retention factor, or RF).

Equation 3: Determination of Inhalation Dose

Inhalation dose = OEL × Inhalation

volume × RF = OEL (mg/m³) × 10 m³ × 0.75

The final step is to compare the calculated skin and inhalation doses and to present the

result as a ratio of skin dose to inhalation dose (the SI ratio). This ratio quantitatively indicates (1) the significance of dermal absorption as a route of occupational exposure to the substance and (2) the contribution of dermal uptake to systemic toxicity. If a substance has an SI ratio greater than or equal to 0.1, it is considered a skin absorption hazard.

Calculation

Table A1 summarizes the data applied in the previously described equations to determine the SI ratio for aniline. The calculated SI ratio was 2.28. On the basis of these results, aniline is not predicted to represent a skin absorption hazard.

Appendix References

- NIOSH [2005]. NIOSH pocket guide to chemical hazards. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2005–149, http://www.cdc.gov/niosh/npg/. Accessed: 02-10-15.
- NIOSH [2009]. Current intelligence bulletin 61: a strategy for assigning new NIOSH skin notations. Cincinnati, OH: U.S. Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, DHHS (NIOSH) Publication No. 2009-147, http://www.cdc.gov/niosh/docs/2009-147/ pdfs/2009-147.pdf. Accessed: 02-10-15.
- SRC [ND]. Interactive PhysProp database demo, http://esc.syrres.com/fatepointer/webprop. asp?CAS=62533. Accessed: 02-10-15.

Variables used in calculation	Units	Value
Skin permeation coefficient		
Permeation coefficient of stratum corneum lipid path(k_{psc})	cm/hr	0.00316
Permeation coefficient of the protein fraction of the stra- tum corneum (k_{pol})	cm/hr	1.5743 × 10 ⁻⁵
Permeation coefficient of the watery epidermal layer (k_{aq})	cm/hr	0.2591
Molecular weight $(MW)^*$	amu	93.1
Base-10 logarithm of its octanol–water partition coefficient (Log K_{ow})*	None	0.9
Calculated skin permeation coefficient (k_p)	cm/hr	0.00314
Skin dose		
Water solubility $(S_w)^*$	mg/cm ³	36
Calculated skin permeation coefficient (k_{p})	cm/hr	0.00314
Estimated skin surface area (palms of hand)	cm ²	360
Exposure time	hr	8
Calculated skin dose	mg	325.1
Inhalation Dose		
Occupational exposure limit (OEL) ⁺	mg/m ³	19
Inhalation volume	m ³	10
Retention factor (RF)	None	0.75
Inhalation dose	mg	142.5
Skin dose–to–inhalation dose (SI) ratio	None	2.28

Table A1. Summary of data used to calculate the SI ratio for aniline

*Variables identified from SRC [ND]. *The OEL used in calculation of the SI ratio for aniline was the NIOSH recommended exposure limit (REL) [NIOSH 2005].

This page intentionally left blank.



Delivering on the Nation's promise: safety and health at work for all people through research and prevention

To receive NIOSH documents or more information about occupational safety and health topics, contact NIOSH at

1–800–CDC–INFO (1–800–232–4636) TTY: 1–888–232–6348 CDC-INFO: www.cdc.gov/info

or visit the NIOSH website at **www.cdc.gov/niosh**.

For a monthly update on news at NIOSH, subscribe to *NIOSH eNews* by visiting **www.cdc.gov/niosh/eNews**.

DHHS (NIOSH) Publication No. 2015–192

SAFER • HEALTHIER • PEOPLE[™]