



Transport of Pollutants Through Liquid-Gas Interfaces – a Numerical Approach

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NIOSH-ERC

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OUTLINE

- **Motivation**
- **Objectives of Present Study**
- **Problem Description**
- **Steps in Solution Procedure**
- **Results**
- **Future Work**

MOTIVATION

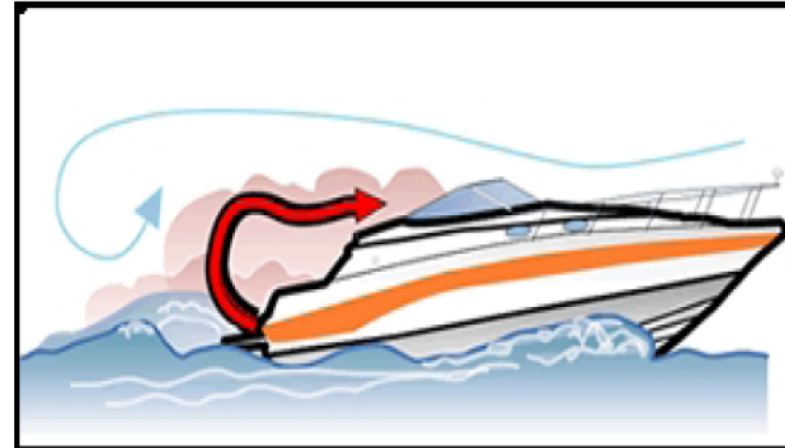
MOTIVATION:

- Carbon Monoxide (CO) is ejected in the exhaust of combustion-based systems, e.g., watercraft.
- Being lighter than water, CO rises to water surface behind boat,
- Gets entrained into the cabin due to “Station-Wagon effect”.
- Tests conducted by NIOSH show high CO concentration in back deck and cabin.
- Marine workers are exposed to high concentration of CO – higher than Immediately Dangerous to Life and Health (IDLH) value of 1200 ppm.

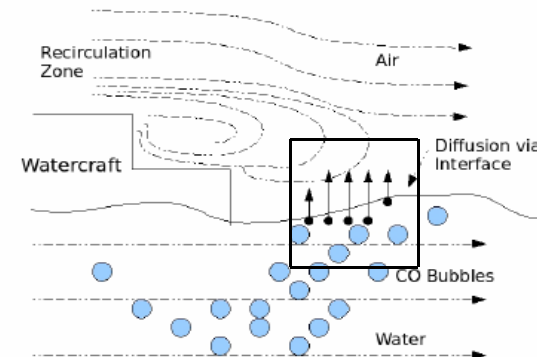
PREVIOUS RESEARCH CONDUCTED AT CFDRL:

- Studied dispersal of exhaust gases under the hull of a watercraft by modeling the water surface as a slip boundary.
- Effect of free surface on exhaust dispersal was not considered.

Station-Wagon effect, *Courtesy: www.boat-ed.com*

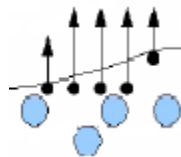


Transport of CO in water and across water-air interface



OBJECTIVES

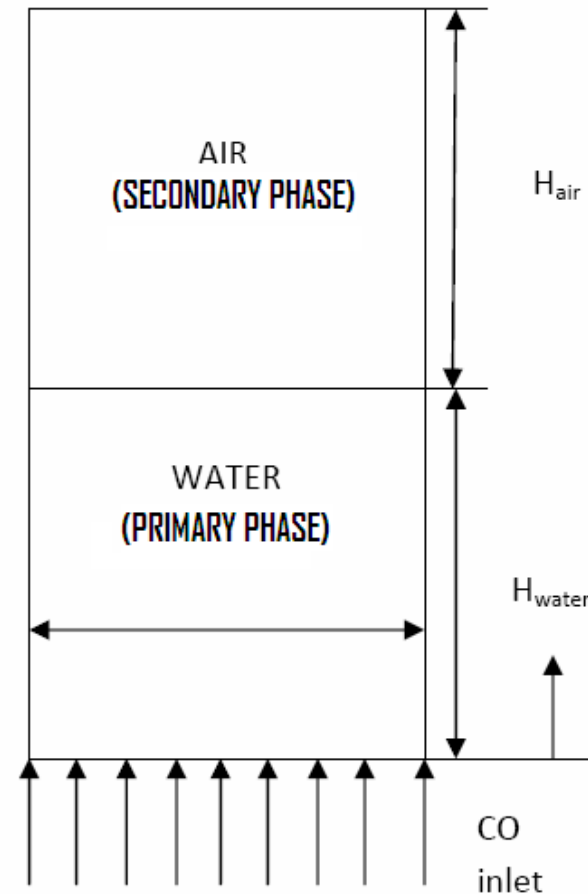
- Understand the flow physics involved in the transport of pollutants through an air-water interface.
- Develop a Computational Fluid Dynamics (CFD) model to simulate this flow physics.
- Accurately predict pollutant transport into air above the water surface by considering a unit process involving the pollutants and the air-water interface.



PROBLEM DESCRIPTION

- Cylindrical vertical column:
Height: 2m
Diameter: 0.2m.
- Static height of water column:
 H_{water} : 1m.
- Mixture composition at inlet:
Volume Fraction of CO: 0.5
Volume Fraction of Water: 0.5
- CO bubbles, of diameter of 0.5 cm, flow in through the bottom, with a velocity of 0.1 m/s.
- Atmospheric pressure at top of air column.

Schematic of simulation domain



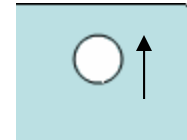
STEPS IN SOLUTION PROCEDURE

- Unsteady Incompressible Navier-Stokes equations are solved.
- Eulerian Multiphase model is used, treating phases as interpenetrating continua.

Primary phase: Water

Secondary phases: Air and CO

- “Virtual mass effect” term included in the momentum equations for both phases – accounts for inertia of primary-phase mass encountered by the accelerating bubbles.



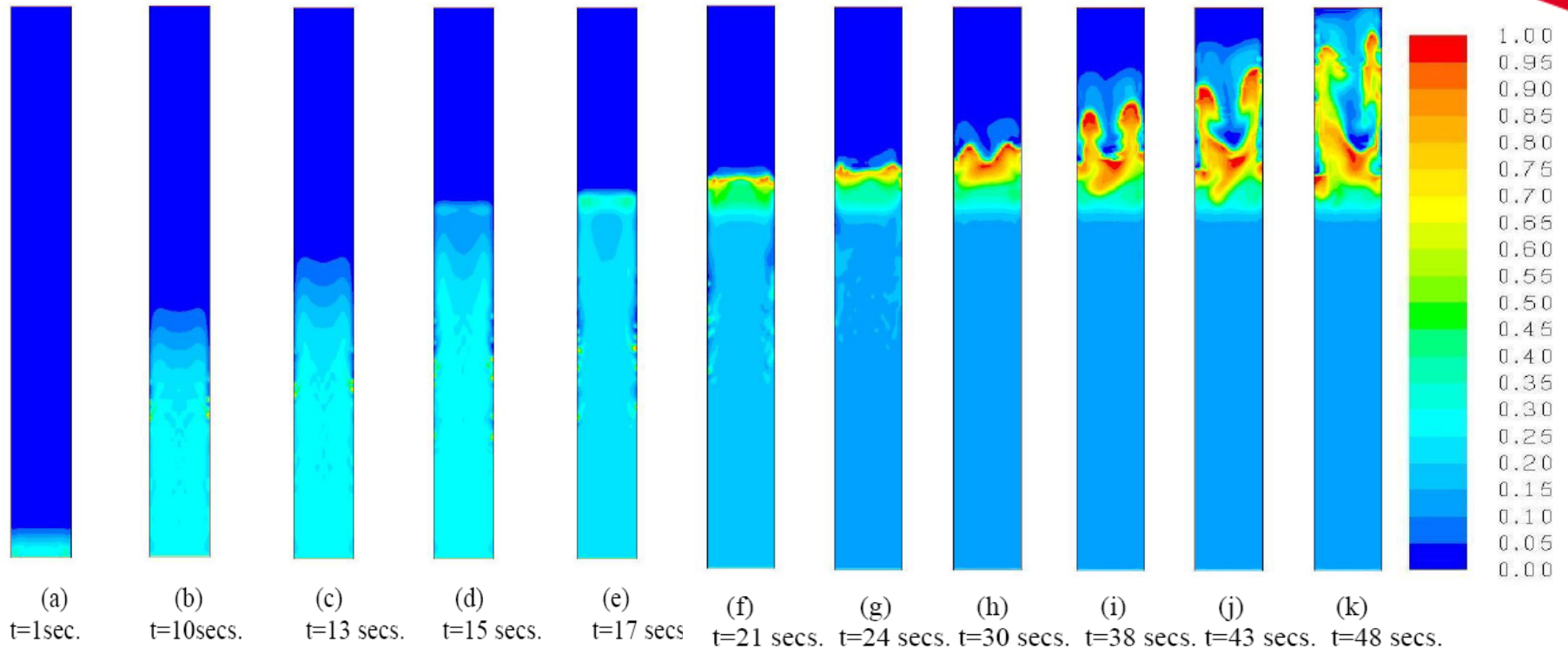
- Schiller-Naumann drag model is used. In definition of C_D , the Reynolds number uses the magnitude of the velocity of the primary phase relative to that of the secondary phase. Acceptable for use with all fluid-fluid pairs.
- Turbulence is modeled using the mixture k - ϵ equations.

STEPS IN SOLUTION PROCEDURE (Cont'd)

- Pressure-velocity coupling is achieved using the Phase-coupled SIMPLE algorithm.
 - Velocities are solved coupled by phases, but in a segregated fashion.
 - Pressure correction based on *total volume continuity*.
- Second-order upwind scheme is used to discretize convection terms.
- The finite-volume flow solver FLUENT is used.
- Grid in r and z directions: 30x300.
- Time step: 0.0001 seconds.
- The solution at each time-step is assumed converged when scaled residuals of continuity and momentum conservation equations reach the order of 10^{-3} .
- Simulation is continued until CO exits the column, i.e., for approximately 48 seconds.

RESULTS

Transient CO motion along the vertical column



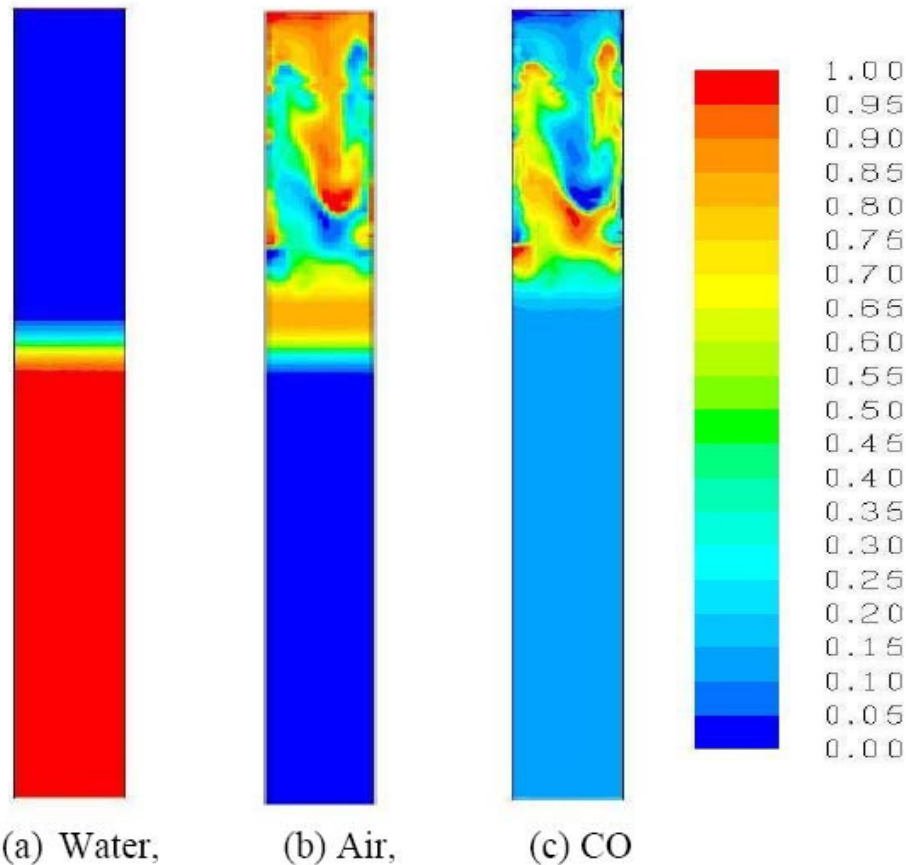
Key observations about CO behavior:

- CO gas enters in the form of bubbles, and rises to air-water interface.
- CO gas traverses water domain in approx. 15 secs to reach the air-water interface, owing to substantial driving force.
- Slow dispersal of CO in the air; CO takes approx. 35 secs to reach top of air column.

RESULTS (Cont'd)

General Observations about Air, Water and CO Volume Fractions

Volume Fraction Contours at t=48 seconds.



- Rapid CO dispersal in water domain:
 - *Large density difference* : CO is 1000 times lighter than water.
 - *Initial momentum.*
- Fractional increase in the level of the air-water interface as CO-water mixture enters the vertical column.
- CO bubbles rupture at air-water interface, and CO gas is freely dispersed into air region.
- Slow dispersal of CO in air domain:
 - *Milder density difference*: CO is only marginally lighter than Air.
 - Loss of initial momentum due to liquid drag.

CONCLUSIONS

- ***Rapid CO movement through water domain – due to initial momentum and large density difference between water and CO gas.***
- ***At the air-water interface – CO bubbles rupture, and CO gas is freely dispersed into the air.***
- ***Milder density difference in the air domain – much slower dispersal rates compared to CO movement in the liquid region.***
- **Air-water interface inadequately resolved.**



FUTURE WORK

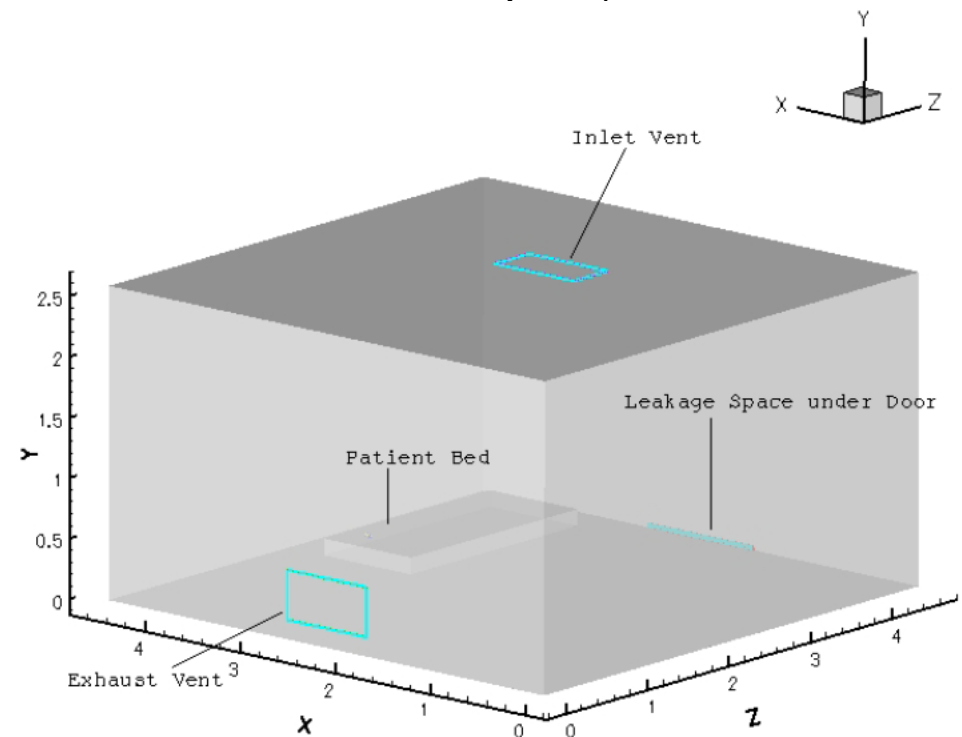
- Enhance *problem geometry* to include a watercraft and its surrounding environment.
- Devise methodology to adequately capture the air-water interface.
- Explore methods to capture *wave motion* at the air-water interface, *and the corresponding effect on the motion of the gas crossing the interface.*

HOW DID PRP HELP US?!

KEY OUTCOMES:

- Developed a deeper understanding of Multi-phase and Multi-species flow phenomena.
- Extended the use of Computational Fluid Dynamics to applications in Environmental Health and Occupational Safety.

Numerical Simulation of Pandemic Flu Dispersal in an Airborne Infection Isolation Room (AIIR)



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Main Menu

Hosted by: The University of Cincinnati Education and Research Center Supported by: The National
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- ◆ **Welcome and Opening Remarks**
- ◆ **Keynote Speakers**
- ◆ **Podium Presentations**
- ◆ **Poster Presentations**
- ◆ **Video Montage of the 10th Annual PRP
Symposium**
- ◆ **Participating Universities**
- ◆ **Steering Committee Members**
- ◆ **Acknowledgements**
- ◆ **Problems Viewing the Videos**

Produced by Kurt Roberts Department of Environmental Health
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