


## Assessing the accuracy of commercially available gas sensors for the measurement of ambient ozone and nitrogen dioxide

Kelechi Isiugo, Nicholas Newman, Roman Jandarov, Sergey A. Grinshpun & Tiina Reponen


To cite this article: Kelechi Isiugo, Nicholas Newman, Roman Jandarov, Sergey A. Grinshpun & Tiina Reponen (2018) Assessing the accuracy of commercially available gas sensors for the measurement of ambient ozone and nitrogen dioxide, Journal of Occupational and Environmental Hygiene, 15:11, 782-791, DOI: [10.1080/15459624.2018.1513135](https://doi.org/10.1080/15459624.2018.1513135)

To link to this article: <https://doi.org/10.1080/15459624.2018.1513135>

 View supplementary material 

 Accepted author version posted online: 29 Aug 2018.  
Published online: 29 Nov 2018.

 Submit your article to this journal 

 Article views: 244

 View Crossmark data 



## Assessing the accuracy of commercially available gas sensors for the measurement of ambient ozone and nitrogen dioxide

Kelechi Isiugo<sup>a</sup>, Nicholas Newman<sup>a,b</sup>, Roman Jandarov<sup>a</sup>, Sergey A. Grinshpun<sup>a</sup>, and Tiina Reponen<sup>a</sup>

<sup>a</sup>Department of Environmental Health, University of Cincinnati, Cincinnati, Ohio, USA; <sup>b</sup>Cincinnati Children's Hospital Medical Center, Cincinnati, Ohio, USA

### ABSTRACT

The objective of the National Institute for Occupational Safety and Health (NIOSH) accuracy criterion is to ensure that measurements from monitoring devices are within  $\pm 25\%$  of the true concentration of the analyte with 95% certainty. To determine whether NO<sub>2</sub> and O<sub>3</sub> sensors meet this criterion, three commercially available units (Cairclip O<sub>3</sub>/NO<sub>2</sub>, Aeroqual NO<sub>2</sub>, and Aeroqual O<sub>3</sub> sensors) were co-located three times with validated instruments (NO<sub>x</sub> chemiluminescence [NO<sub>2mon</sub>] and photometric O<sub>3</sub> analyzers [O<sub>3mon</sub>]) at an outdoor monitoring station. As cofactors of sensor performance such as temperature (T) and relative humidity (RH) potentially influence the response of NO<sub>2</sub> and O<sub>3</sub> sensors, corrections for cofactors were made by using T, RH, and the sensor measurements to predict measurements made by NO<sub>2mon</sub> and O<sub>3mon</sub> during the first co-location period (training dataset). The developed models were tested in the merged data obtained from the second and third co-location periods (testing dataset). In the training and testing datasets, the mean NO<sub>2</sub> as measured by NO<sub>2mon</sub> was 4.6 ppb (range = 0.4–35 ppb) and 9.4 ppb (range = 1–37 ppb), respectively. The mean O<sub>3</sub> in the training and testing datasets as measured by O<sub>3mon</sub> was 38.8 ppb (range = 1–65 ppb) and 35.7 ppb (range = 1–61 ppb), respectively. None of the sensor measurements in the training dataset were within the NIOSH accuracy criterion (mean error  $\geq 25\%$ ). After correcting for cofactors of sensor performance, the accuracy of the Cairclip O<sub>3</sub>/NO<sub>2</sub> and the Aeroqual O<sub>3</sub> sensors considerably improved when tested with the testing dataset (mean error = -1% and 14%, respectively). However, the Aeroqual NO<sub>2</sub> sensor had an error that was not within  $\pm 25\%$ . Raw measurements from the tested sensors may be unsuitable for assessing workers' exposure to NO<sub>2</sub> and O<sub>3</sub>. Corrections for cofactors of Cairclip O<sub>3</sub>/NO<sub>2</sub> and Aeroqual O<sub>3</sub> sensor performance are required for more accurate occupational exposure assessment.

### KEYWORDS

Aeroqual; Cairclip; direct-reading instruments; low-cost sensors

### Introduction

Nitrogen dioxide (NO<sub>2</sub>) and ground-level ozone (O<sub>3</sub>) are common air pollutants associated with adverse respiratory health effects.<sup>[1–8]</sup> Due to the emission of NO<sub>2</sub> from vehicular exhausts,<sup>[9]</sup> NO<sub>2</sub> is regarded as a traffic-related air pollutant, whereas O<sub>3</sub> is formed through photochemical reactions involving NO<sub>x</sub> (nitrogen oxides) and volatile organic compounds,<sup>[10,11]</sup> which are emitted from vehicular exhausts. Consequently, traffic control workers may have elevated levels of exposure to NO<sub>2</sub> and O<sub>3</sub>, which may be associated with lower lung function and increased respiratory symptoms.<sup>[12–14]</sup> As a result of emissions from the exhausts of internal combustion engines, other outdoor workers such as toll booth workers,<sup>[15]</sup> commercial drivers,<sup>[13,16]</sup> petrol-pump workers,<sup>[17]</sup> air cargo handlers,<sup>[18]</sup> tunnel construction

workers,<sup>[19]</sup> and street cleaners<sup>[20,21]</sup> may also be exposed to higher levels of NO<sub>2</sub> and O<sub>3</sub> than the general population. The lowest occupational exposure limit (OEL) for NO<sub>2</sub> and O<sub>3</sub> are the threshold limit values (TLVs): 200 ppb for NO<sub>2</sub>,<sup>[22]</sup> and 50–200 ppb for O<sub>3</sub> (depending on working conditions).<sup>[23]</sup> In most typical outdoor scenarios, outdoor workers in countries that have enforceable regulatory ambient air quality standards are likely to have occupational exposure to outdoor NO<sub>2</sub> and O<sub>3</sub> below OEL. However, some indoor workers in developed countries and indoor/outdoor workers in developing countries can have NO<sub>2</sub> and O<sub>3</sub> exposures that exceed OELs. Examples are indoor ice skating workers in rinks that use petroleum-based fuels in the resurfacers to maintain ice<sup>[24,25]</sup> and accidental release of O<sub>3</sub> in pulp mills.<sup>[26]</sup> Thus, reliable NO<sub>2</sub> and O<sub>3</sub> monitoring devices are needed for occupational exposure assessment.

In contrast to traditional passive samplers for NO<sub>2</sub> and O<sub>3</sub>,<sup>[27,28]</sup> electronic gas sensors are relatively new direct reading instruments that are used for assessing exposure to NO<sub>2</sub> and O<sub>3</sub>. They are sometimes referred to as “low-cost sensors” because of their relatively affordable price (<\$1,500). Researchers have questioned the accuracy of NO<sub>2</sub> and O<sub>3</sub> sensors, mainly because cofactors such as reactive pollutants, temperature, and relative humidity can influence the response of the sensors to NO<sub>2</sub> and O<sub>3</sub> concentrations.<sup>[29,30]</sup> Due to the concern about the quality of measurements from NO<sub>2</sub> and O<sub>3</sub> sensors, scientists have investigated their accuracy by co-locating them with validated instruments and utilizing Pearson correlation coefficients (*r*) and coefficients of determination (*R*<sup>2</sup>) to assess the quality of the data.<sup>[31–33]</sup> To improve the correlation, the impact of T, RH, and reactive pollutants on sensor performance have been modeled. Applying these models, Cross et al.<sup>[32]</sup> and Zimmerman et al.<sup>[34]</sup> reported an increase in the correlation between corrected sensor measurements and measurements from validated instruments. Nevertheless, computed *r* and *R*<sup>2</sup> based on normality assumption may be inappropriate when the distribution of the analyte is not normally distributed.<sup>[35]</sup> According to scientists at the National Institute for Occupational Safety and Health (NIOSH), direct reading instruments for gaseous exposure assessment should measure within ±25% of the true concentration of the target analyte with 95% certainty.<sup>[36]</sup> This assessment corresponds to the 95% confidence interval (95% CI) accuracy level. To the best of our knowledge, researchers have not demonstrated that NO<sub>2</sub> and O<sub>3</sub> gas sensors meet this accuracy criterion in the field.

The United States Environmental Protection Agency (EPA) performs regulatory actions for ambient NO<sub>2</sub> and O<sub>3</sub> concentrations through Ambient Air Quality Standards.<sup>[37,38]</sup> EPA judges the accuracy of NO<sub>2</sub> and O<sub>3</sub> measuring devices by their performance in detecting NO<sub>2</sub> and O<sub>3</sub> after known concentrations are released onto the measuring devices. The EPA acceptable measurement of ambient NO<sub>2</sub> has a threshold of 15% coefficient of variation and bias, while that of O<sub>3</sub> has a threshold of 7% coefficient of variation and bias.<sup>[39]</sup> The EPA has a list of designated reference and equivalent instruments that can be used to achieve EPA acceptable measurements of ambient NO<sub>2</sub> and O<sub>3</sub>.<sup>[40]</sup> This list includes NO<sub>x</sub> chemiluminescence analyzers as a method for measuring ambient NO<sub>2</sub><sup>[40]</sup> and photometric O<sub>3</sub> analyzers as a method for measuring ambient O<sub>3</sub>.<sup>[40,41]</sup> However, NO<sub>2</sub> and O<sub>3</sub> sensors are not included in the list as reference/

equivalent instruments. A non-exhaustive list of commercially available NO<sub>2</sub> and O<sub>3</sub> sensors includes: Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor,<sup>[42]</sup> Aeroqual GSE NO<sub>2</sub> sensor,<sup>[43]</sup> Aeroqual GSS O<sub>3</sub> sensor,<sup>[44]</sup> MSA NO<sub>2</sub> detector,<sup>[45]</sup> Drager NO<sub>2</sub> sensor,<sup>[46]</sup> and Drager O<sub>3</sub> sensor.<sup>[46]</sup> The Aeroqual O<sub>3</sub> sensor is a metal-oxide-semiconductor ozone sensor,<sup>[44]</sup> whereas the other sensors are electrochemical sensors. Documentation from the sensor indicates that the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor is capable of measuring the sum of NO<sub>2</sub> and O<sub>3</sub>,<sup>[42]</sup> whereas the other sensors named are capable of measuring only their target analyte.<sup>[43–46]</sup> Based on the theory of operation presented in the scientific literature, electrochemical sensors contain electrodes which detect the current produced from a target gas undergoing a reaction in the sensor.<sup>[47,48]</sup> Metal-oxide-semiconductor gas sensors have semiconductors that detect a change in electrical resistance caused by the reaction of a sampled target gas with reducing or oxidizing gases present on the surface of the semiconductor.<sup>[49,50]</sup> There is an inherent lack of sensitivity of target gases in metal-oxide semiconductor sensors.<sup>[51]</sup>

As gas sensors are used for the same purpose as reference/equivalent instruments (i.e., to measure air pollutants relevant to health), this study examined the accuracy of three commercial NO<sub>2</sub> and O<sub>3</sub> gas sensors in the field to validate their use for occupational exposure assessment using the NIOSH accuracy criterion. Additionally, the study examined the correlation of measurements made with the gas sensors and reference/equivalent instruments.

## Methods

### *Investigating the accuracy of NO<sub>2</sub> and O<sub>3</sub> sensors*

One of each sensor [the NO<sub>2</sub> sensor (GSE, s500, 0–1 ppm, Aeroqual), O<sub>3</sub> sensor (GSS, s500, 0–0.15 ppm, Aeroqual) and the O<sub>3</sub>+NO<sub>2</sub> sensor (O<sub>3</sub>/NO<sub>2</sub>, Cairclip, Cairpol)] was co-located outdoors three times side-by-side with the Southwest Ohio Air Quality Agency monitoring station (250 William Howard Taft Rd., Cincinnati, OH). All the sensors were factory calibrated by the manufacturers, and zero calibration was performed on the Aeroqual sensors by the study team. The factory calibration of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor is sufficient for 1 year provided that its operational conditions are adhered to.<sup>[42]</sup> Prior to the commencement of the current study, the sensors had been used for approximately 5 months since the receipt from the manufacturers. They were within the recommended 1-year factory calibration cycle given by the manufacturers. The first co-locating period was between 3 pm

on 07/19/2017 to 2 pm on 07/24/2017, and the second and third co-locating periods were between 3 pm on 08/30/2017 to 2 pm on 09/01/2017 and 3 pm on 10/17/2017 to 2 pm on 10/19/2017, respectively. The sensors were deployed to compare the readings obtained from them to those of the EPA reference/equivalent instruments at the monitoring station (Figure S1). The monitoring station uses a NO<sub>x</sub> chemiluminescence analyzer (Model T200, Teledyne) [NO<sub>2mon</sub>] and a photometric O<sub>3</sub> analyzer (Model 400E, Teledyne) [O<sub>3mon</sub>]. The recommended operating conditions of

Data from all measurements were imported into R Studio<sup>[55]</sup> for data analysis. Only hourly averaged measurements  $\geq$  the LOD of the sensors and reference instruments were used in this study. First, accuracy was determined by estimating the error associated with measurements of the NO<sub>2</sub> and O<sub>3</sub> sensors. This was achieved by creating variables that contained the percentage difference between the measurements made by the sensors and measurements from the reference instruments (Equation (1)):

Second, the mean error was calculated and a bias-cor-

$$Error = \left( \frac{Sensor\ measurements_i - Reference\ instrument\ measurements_i}{Reference\ instrument\ measurements_i} \right) \times 100, \quad (1)$$

where  $i$  = individual observation in the dataset.

the measuring devices are  $-20$ – $40^\circ\text{C}$  and 10–90% RH for the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor,  $0$ – $40^\circ\text{C}$  and 15–90% RH for the Aeroqual NO<sub>2</sub> sensor, and  $0$ – $40^\circ\text{C}$  and 10–90% RH for the Aeroqual O<sub>3</sub> sensor,  $5$ – $40^\circ\text{C}$  and 0–95% RH for the NO<sub>x</sub> chemiluminescence analyzer,  $5$ – $40^\circ\text{C}$  and 0–90% RH for the photometric O<sub>3</sub> analyzer.<sup>[43,44,52–54]</sup> Data from the NO<sub>x</sub> chemiluminescence analyzer and photometric O<sub>3</sub> analyzer were added together to obtain “reference O<sub>3</sub>+NO<sub>2</sub>.” The limits of detection (LOD) of the measuring devices were obtained from their manuals (20 ppb for the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor, 5 ppb for the Aeroqual NO<sub>2</sub> sensor, 1 ppb for the O<sub>3</sub> sensor, 0.4 ppb for the NO<sub>x</sub> chemiluminescence analyzer and 0.6 ppb for the photometric O<sub>3</sub> analyzer).<sup>[43,44,52–54]</sup>

Temperature (T) and relative humidity (RH) were simultaneously measured with a temperature/relative humidity monitor (O83E, Met One Instruments, Inc.). The time and date on all devices were synchronized and were set to record measurements every minute. For quality assurance, data from the reference/equivalent instruments were reviewed by the operators of the monitoring station (Southwest Ohio Air Quality Agency). At the end of the sampling period, hourly averages of NO<sub>2</sub>, O<sub>3</sub>, T, and RH were provided by the operators of the monitoring station.

All the data logged per minute by the sensors (including data  $<$  LOD) were downloaded to a computer using the manufacturers’ proprietary software, and hourly averages of NO<sub>2</sub> and O<sub>3</sub> levels measured by the sensors were calculated. Data obtained from the reference/equivalent instruments at the monitoring station were regarded as the “gold standards” providing the true concentration of ambient NO<sub>2</sub> and O<sub>3</sub>.

rected and accelerated (BCa) bootstrap method was used to calculate 95% CI for the mean error.<sup>[56–59]</sup> Bootstrap was employed for the calculation of 95% CI because the created error variables were non-normal. Accuracy was defined as having a 95% CI of mean error (lower 95% CI of mean error  $<$  true mean error  $<$  upper 95% CI of mean error) within  $-25$  to  $+25\%$ .

The correlation of measurements from the sensors and reference instruments were also calculated in order to make results from the current study comparable to results of existing studies. The following tests for accuracy and correlation of the sensor measurements compared to the O<sub>3</sub> reference-equivalent instrument (O<sub>3mon</sub>), NO<sub>2</sub> reference instrument (NO<sub>2mon</sub>) and (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub> were carried out:

- Aeroqual O<sub>3</sub> sensor and O<sub>3mon</sub>;
- Aeroqual NO<sub>2</sub> sensor and NO<sub>2mon</sub>; and
- Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor and (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>.

### Developing corrective models

Data from the first co-locating period were utilized to develop regression models to correct for measured cofactors of sensor performance that potentially interfere with the accuracy of the sensors. The data collected during this period is described as the “Training Dataset.” To develop a model to correct for measured cofactors of the Aeroqual O<sub>3</sub> sensor performance, and to correct for the inherent error in the Aeroqual O<sub>3</sub> sensor, measurements of T, RH, NO<sub>2mon</sub>, and the Aeroqual O<sub>3</sub> sensor were used to predict O<sub>3mon</sub> (Equation (S1) is available in online supplemental materials). Due to the nonlinear

**Table 1.** Summary of measurements in the training dataset.

	T (°C)	RH (%)	O <sub>3mon</sub> (ppb)	Aeroqual O <sub>3</sub> (ppb)	NO <sub>2mon</sub> (ppb)	Aeroqual NO <sub>2</sub> (ppb)	(O <sub>3</sub> +NO <sub>2</sub> ) <sub>mon</sub> (ppb)	Cairclip O <sub>3</sub> /NO <sub>2</sub> (ppb)
Minimum	21.9	39.0	1.0	1.7	0.4	7.6	21.0	31.6
Maximum	34.5	90.0	65.0	69.7	35.0	92.9	69.0	187.6
Mean	27.4	69.8	38.8	43.1	4.6	30.5	43.4	70.0
SD	4.0	13.5	14.5	14.9	6.0	11.3	11.4	24.5
n	120	120	120	107	120	120	120	120

Measurements are based on hourly averages, training dataset = data used to develop the regression models. O<sub>3mon</sub> = O<sub>3</sub> measured by the EPA reference-equivalent instrument with a photometric O<sub>3</sub> analyzer, Aeroqual O<sub>3</sub> = O<sub>3</sub> measured by the Aeroqual GSS O<sub>3</sub> sensor, NO<sub>2mon</sub> = NO<sub>2</sub> measured by the EPA reference instrument with a NO<sub>x</sub> chemiluminescence analyzer, Aeroqual NO<sub>2</sub> = NO<sub>2</sub> measured by the Aeroqual GSE NO<sub>2</sub> sensor, (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub> = the sum of NO<sub>2</sub> and O<sub>3</sub> measured by the EPA reference instruments, Cairclip O<sub>3</sub>/NO<sub>2</sub> = O<sub>3</sub>+NO<sub>2</sub> measured by the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor, n = number of observations.

response of metal-oxide semiconductor sensors to the target gas and cofactors of sensor response,<sup>[51]</sup> quadratic terms were added to the independent variables. Variable selection was performed with a stepwise regression model using the minimum Akaike Information Criterion (AIC) for model selection. To develop a model to correct the effect of measured cofactors of the Aeroqual NO<sub>2</sub> sensor performance, and to correct for the inherent error in the Aeroqual NO<sub>2</sub> sensor, measurements of T, RH, O<sub>3mon</sub>, and the Aeroqual NO<sub>2</sub> sensor were used to predict NO<sub>2mon</sub> (Equation (S2)). Last, to develop a model to correct the effect of measured cofactors of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor performance, and to correct for the inherent error in the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor, measurements of T, RH, and the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor were used to predict (O<sub>3</sub> + NO<sub>2</sub>)<sub>mon</sub> (Equation (S3)). After developing the regression models as described, the independent variables that were not significant ( $P > 0.05$ ) were removed from the models, and the final models were used to correct the raw sensor data in the subsequent field trial.

### Testing the models developed

The sensors were co-located at the monitoring station twice to test whether the accuracy of the sensor measurements could be improved by utilizing the regression coefficients obtained from the final models. The data obtained to test the models (testing dataset) were collected from the second and third co-locating periods. These data were merged together to form one testing dataset. Measurements from the testing dataset were multiplied by the corresponding coefficients obtained from the final models.

Errors of the corrected sensor measurements were calculated as described in Equation (1), and mean errors and 95% CI were calculated. In addition, the  $R^2$  of the measurements from the reference instruments and the corrected sensor measurements were calculated.

## Results

### Summary of measurements

In the training dataset, 10.8% of hourly averaged measurements made by the Aeroqual O<sub>3</sub> sensor were < its LOD of 1 ppb. In the testing dataset, 42.9% of hourly averaged measurements made by the Aeroqual NO<sub>2</sub> sensor were < its LOD of 5 ppb, and 34.1% of hourly averaged measurements from the Aeroqual O<sub>3</sub> sensor were < its LOD of 1 ppb. Other sensor measurements were within their LOD. Hourly averaged data < LOD were removed.

Tables 1 and 2 present the summary statistics of hourly averaged NO<sub>2</sub>, O<sub>3</sub>, O<sub>3</sub>+NO<sub>2</sub>, T, and RH from the reference instruments and sensors in the training and testing datasets. In the training dataset (i.e., the dataset used to develop the regression models), ambient T ranged from 21.9–34.5°C, and ambient RH ranged from 39–90%. In the testing dataset (i.e., the dataset used to test the developed regression models), ambient T ranged from 7.4–28.4°C, and ambient RH ranged from 35–87%. The T and RH values were within the operating conditions specified by the manufacturers. The mean NO<sub>2</sub> as measured by NO<sub>2mon</sub> was 4.6 ppb (range = 0.4–35 ppb) in the training dataset and 9.4 ppb (range = 1–37 ppb) in the testing dataset. Furthermore, the mean O<sub>3</sub> concentration as measured by O<sub>3mon</sub> was 38.8 ppb (range = 1–65 ppb) in the training dataset and 35.7 ppb (range = 1–61 ppb) in the testing dataset.

### Accuracy and correlation during the first co-location period

Table 3 and Figure S2 present the results obtained from the first co-location of the sensors with reference instruments at the monitoring station. During the first co-locating period, when the sensors were deployed to obtain data for the corrective models, there was a consistent positive bias in sensor measurements. Results of raw measurements from the Aeroqual O<sub>3</sub> sensor when compared to O<sub>3mon</sub> had a mean error of 30%



**Table 2.** Summary of measurements in the testing dataset.

	T (°C)	RH (%)	O <sub>3mon</sub> (ppb)	Aeroqual O <sub>3</sub> (ppb)	NO <sub>2mon</sub> (ppb)	Aeroqual NO <sub>2</sub> (ppb)	(O <sub>3</sub> +NO <sub>2</sub> ) <sub>mon</sub> (ppb)	Cairclip O <sub>3</sub> /NO <sub>2</sub> (ppb)
Minimum	7.4	35.0	1.0	1.4	1.0	5.2	21.0	21.9
Maximum	28.4	87.0	61.0	63.1	37.0	31.2	68.0	86.1
Mean	17.5	64.5	35.7	33.1	9.4	15.0	45.1	47.3
SD	5.8	15.2	14.9	17.6	7.7	7.6	12.0	12.8
n	91	91	91	60	91	52	91	91

Measurements are based on hourly averages, testing dataset = data used to test the developed regression models. See footnotes of Table 1.

**Table 3.** Accuracy and correlation of raw data from the NO<sub>2</sub> and O<sub>3</sub> sensors obtained from the training dataset (data used to develop the regression models).

s/n	Comparison	n	R <sup>2</sup>	Mean error	95% CI of mean error	Within NIOSH accuracy criterion
A	Aeroqual O <sub>3</sub> vs. O <sub>3mon</sub>	107	0.71	30%	13–78%	No
B	Aeroqual NO <sub>2</sub> vs. NO <sub>2mon</sub>	120	0.03	4264%	3160–5662%	No
C	Cairclip O <sub>3</sub> /NO <sub>2</sub> vs. (O <sub>3</sub> +NO <sub>2</sub> ) <sub>mon</sub>	120	0.32	65%	58–76%	No

n = number of observations, R<sup>2</sup> = coefficient of determination based on linear regression, mean error = percent difference between the measurements made by the sensors and measurements from the reference instruments (equation 1), 95% CI of mean error was obtained from BCa bootstrap confidence intervals, NIOSH accuracy criterion refers to 95% CI estimated mean error of  $\pm 25\%$ , Aeroqual O<sub>3</sub> = O<sub>3</sub> measured by the Aeroqual GSS O<sub>3</sub> sensor, O<sub>3mon</sub> = O<sub>3</sub> measured by the EPA reference-equivalent instrument with a photometric O<sub>3</sub> analyzer, Aeroqual NO<sub>2</sub> = NO<sub>2</sub> measured by the Aeroqual GSE NO<sub>2</sub> sensor, NO<sub>2mon</sub> = NO<sub>2</sub> measured by the EPA reference instrument with a NO<sub>x</sub> chemiluminescence analyzer, Cairclip O<sub>3</sub>/NO<sub>2</sub> = O<sub>3</sub>+NO<sub>2</sub> measured by the Cairclip O<sub>3</sub>/NO<sub>2</sub> Sensor, (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub> = the sum of NO<sub>2</sub> and O<sub>3</sub> measured by the EPA reference instruments.

(95% CI = 13, 78) (Table 3, Row A, and Figure S2 A and B). The variation in raw measurements of the Aeroqual O<sub>3</sub> sensor explained 71% of the variation in O<sub>3mon</sub>.

Raw measurements from the Aeroqual NO<sub>2</sub> sensor when compared to NO<sub>2mon</sub> had a mean error of 4264% (95% CI = 3160, 5662) (Table 3, Row B, and Figure S2 C and D). The variation in raw measurements of the Aeroqual NO<sub>2</sub> sensor explained only 3% of the variation in NO<sub>2mon</sub>. Raw measurements from the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor when compared to (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub> had a mean error of 65% (95% CI = 58, 76) (Table 3, Row C, and Figure S2 E and F). Furthermore, the variation of raw measurements of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor explained 32% of the variation in (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>.

It appeared that higher RH and lower T were associated with lower measurements of O<sub>3</sub> and NO<sub>2</sub> (Figure S2). However, results of the variable selection for the final model showed that the effect of T on NO<sub>2mon</sub> or O<sub>3mon</sub> was not significant (results not shown). Consequently, T was not included in the final models (Table 4).

The final model for correcting the measured cofactors of the Aeroqual O<sub>3</sub> sensor performance included the following independent variables: The Aeroqual O<sub>3</sub> sensor squared, NO<sub>2mon</sub> and RH<sup>2</sup> (relative humidity<sup>2</sup>) (Table 4A). In the model developed, measurements of Aeroqual O<sub>3</sub> sensor and RH had a nonlinear relationship with O<sub>3mon</sub> (Table 4A).

The final model for correcting the measured cofactors of the Aeroqual NO<sub>2</sub> sensor performance

included the measurements of the Aeroqual NO<sub>2</sub> sensor, O<sub>3mon</sub>, and RH as independent variables (Table 4B). One ppb increase in NO<sub>2</sub> measured by the Aeroqual NO<sub>2</sub> sensor was associated with 0.1 ppb increase in NO<sub>2mon</sub> (Table 4B).

Last, the final model for correcting the measured cofactors of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor performance included the measurements of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor and RH as independent variables (Table 4C). One ppb increase in O<sub>3</sub>+NO<sub>2</sub> measured by the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor was associated with 0.1 ppb increase in (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub> (Table 4C).

### Accuracy and correlation in the testing dataset and obtaining corrected readings from the sensors

Table 5 and Figures S3–S5 presents the results obtained from co-locating the sensors with reference instruments at the monitoring station in the testing dataset. Prior to the correction of the Aeroqual O<sub>3</sub> sensor, raw measurements of the Aeroqual O<sub>3</sub> sensor when assessed for accuracy against O<sub>3mon</sub>, had a mean error of -9% (95% CI = -18, 3) in the testing dataset (Table 5, Row A, and Figure S3 A and B). After testing the corrective model that controlled for measured cofactors of the Aeroqual O<sub>3</sub> sensor performance (NO<sub>2</sub> and RH) and the inherent error in the measurements, the corrected measurements of the Aeroqual O<sub>3</sub> sensor when assessed for accuracy against O<sub>3mon</sub>, had a mean error of -1% (95% CI = -14, 9) (Table 5, Row A, and Figure S3 C and D). Prior to the correction, the variation of raw measurements from the

**Table 4.** Results of the models for correcting the effect of measured cofactors of sensor performance.

A. Aeroqual O <sub>3</sub> Sensor <sup>A</sup>			B. Aeroqual NO <sub>2</sub> Sensor <sup>B</sup>			C. Cairclip O <sub>3</sub> /NO <sub>2</sub> Sensor <sup>C</sup>		
Parameters	Regression estimate	P-value	Parameters	Regression estimate	P-value	Parameters	Regression estimate	P-value
Intercept	46.974	< 0.001	Intercept	37.149	< 0.001	Intercept	70.474	< 0.001
(Aeroqual O <sub>3</sub> ) <sup>2</sup>	0.005	< 0.001	Aeroqual NO <sub>2</sub>	0.147	< 0.001	Cairclip O <sub>3</sub> /NO <sub>2</sub>	0.139	< 0.001
NO <sub>2mon</sub>	-0.936	< 0.001	O <sub>3mon</sub>	-0.518	< 0.001	—	—	—
(RH) <sup>2</sup>	-0.003	< 0.001	RH	-0.245	< 0.001	RH	-0.534	< 0.001

<sup>A</sup> response variable = O<sub>3mon</sub>, R<sup>2</sup> = 0.87 <sup>B</sup> response variable = NO<sub>2mon</sub>, R<sup>2</sup> = 0.66 <sup>C</sup> response variable = (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>, R<sup>2</sup> = 0.69

<sup>A</sup> result of equation S1 (n = 107) <sup>B</sup> result of equation S2 (n = 120) <sup>C</sup> result of equation S3 (n = 120)

See Table 3 for other footnotes.

**Table 5.** Accuracy and correlation of corrected NO<sub>2</sub> and O<sub>3</sub> sensor readings obtained from the testing data set (data used to test the regression models).

s/n	Comparison	n	Action	R <sup>2</sup>	Mean error	95% CI of mean error	Within NIOSH accuracy criterion
A.	Aeroqual O <sub>3</sub> vs. O <sub>3mon</sub>	60	Raw	0.74	-9%	-18-3%	Yes
			Testing model in Table 4A	0.80	-1%	-14-9%	Yes
B.	Aeroqual NO <sub>2</sub> vs. NO <sub>2mon</sub>	52	Raw	0.08	150%	98-212%	No
			Testing model in Table 4B	0.24	-30%	-48 - -10%	No
C.	Cairclip O <sub>3</sub> /NO <sub>2</sub> vs. (O <sub>3</sub> +NO <sub>2</sub> ) <sub>mon</sub>	91	Raw	0.63	24%	20-29%	No
			Testing model in Table 4C	0.65	14%	9-18%	Yes

See Table 3 for footnotes.

Aeroqual O<sub>3</sub> sensor explained 74% of the variation of O<sub>3mon</sub> in the testing dataset. After correction, the variation of the corrected Aeroqual O<sub>3</sub> sensor measurements explained 80% of the variation of O<sub>3mon</sub>.

The raw measurements of the Aeroqual NO<sub>2</sub> sensor when assessed for accuracy against NO<sub>2mon</sub>, had a mean error of 150% (95% CI = 98, 212) in the testing dataset (Table 5, Row B, and Figure S4 A and B). After controlling for measured cofactors (O<sub>3</sub> and RH) and the inherent error in the measurements, the corrected measurements of the Aeroqual NO<sub>2</sub> sensor when assessed for accuracy against NO<sub>2mon</sub>, had a mean error of -30% (95% CI = -48, -10) (Table 5, Row B, and Figure S4 C and D). The variation of raw measurements from the Aeroqual NO<sub>2</sub> sensor explained only 8% of the variation of NO<sub>2mon</sub>. After correction, the variation of the corrected Aeroqual NO<sub>2</sub> sensor measurements explained 24% of the variation of NO<sub>2mon</sub>.

The raw measurements of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor when assessed for accuracy against (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>, had a mean error of 24% (95% CI = 20, 29) in the testing dataset (Table 5, Row C, and Figure S5 A and B). After controlling for RH and the inherent error in the sensor measurements, the corrected measurements of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor when assessed for accuracy against (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>, had a mean error of 14% (95% CI = 9, 18) (Table 5, Row D, and Figure S5 C and D). The variation of raw measurements from the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor explained 63% of the variation of (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>. After correction, the variation of the corrected Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor

measurements explained 65% of the variation of (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub>.

## Discussion

Results from this study show that none of the measurements of the three tested sensors, except for the Aeroqual O<sub>3</sub> sensor in the testing dataset, had a 95% CI of mean error within  $\pm 25\%$ . Thus, for the raw sensor measurements, only the Aeroqual O<sub>3</sub> sensor measurements in the testing dataset were within the NIOSH accuracy criterion (95% CI of the mean error = -18-3%). However, in the training dataset, the Aeroqual O<sub>3</sub> sensor had a 95% CI of the mean error ranging from 13-78% (outside  $\pm 25\%$ ). Because 25% is included in the observed 13-78% range, the data show that the accuracy of the Aeroqual O<sub>3</sub> sensor is varied, and may be accurate only under specific conditions. After controlling for NO<sub>2</sub>, RH, and the inherent error in the Aeroqual O<sub>3</sub> sensor measurements, corrected measurements of the Aeroqual O<sub>3</sub> sensor improved in accuracy (mean error = -1%). This finding suggests that utilizing a calibration model that controls the effect of RH and NO<sub>2</sub> on the performance of the Aeroqual O<sub>3</sub> sensor may be required prior its use for occupational exposure assessment. The inaccurate results of the raw sensor measurements may not be due to sensor drift, as the sensors were within the recommended annual factory calibration cycle given by the manufacturers.

Comparison of measurements of the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor with ambient O<sub>3</sub>+NO<sub>2</sub> concentrations

after controlling for RH and the inherent error in the sensor measurements showed accurate results (mean error = 14% [95% CI = 9–18%]). Conversely, measurements of the Aeroqual NO<sub>2</sub> sensor were not accurate, even after controlling for cofactors of the Aeroqual NO<sub>2</sub> sensor performance and the inherent error in measurements. The data suggest that Aeroqual NO<sub>2</sub> sensor may not be an accurate instrument for monitoring outdoor NO<sub>2</sub> exposures.

Because RH was a significant cofactor of the sensors performance and T was as not, the impact of varying levels of RH on the sensors performance is more important than fluctuating T. This indicates that RH needs to be measured in parallel with the Aeroqual O<sub>3</sub> and Cairclip O<sub>3</sub>/NO<sub>2</sub> sensors when they are used for exposure assessment. In addition, simultaneous measurements of NO<sub>2</sub> may be needed when the Aeroqual O<sub>3</sub> sensor is deployed for measurements. As controlling for the effect of varying levels of RH was sufficient enough to make the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor measurements within  $\pm 25\%$  of O<sub>3</sub>+NO<sub>2</sub>, it is possible that the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor measures accurately in environments that have low RH variation.

Our results show unequal variance of the tested sensors when compared to the corresponding reference instruments. In the training dataset, O<sub>3mon</sub> and the Aeroqual O<sub>3</sub> sensor had similar standard deviations [SD] ( $\sim 15$  ppb). However, this was not the case in the testing dataset (O<sub>3mon</sub> SD  $\sim 15$  ppb and Aeroqual O<sub>3</sub> SD  $\sim 18$  ppb). Furthermore, the SD of NO<sub>2mon</sub> and the Aeroqual NO<sub>2</sub> sensor were markedly different in the training dataset, but approximately equal (8 ppb) in the testing dataset. Similarly, the Cairclip O<sub>3</sub>/NO<sub>2</sub> sensor and the reference (O<sub>3</sub>+NO<sub>2</sub>)<sub>mon</sub> instruments had markedly different SD in the training dataset, but similar SD in the testing dataset. The data clearly show that the performance of the sensors is influenced by environmental conditions such as RH, which was controlled in the corrective models. However, it must be noted that other unmeasured cofactors of the sensors performance may exist.

Zimmerman et al. discovered that the response of one electrochemical NO<sub>2</sub> sensor (Alphasense ID: NO<sub>2</sub>-B43F) was influenced by T, RH, CO, CO<sub>2</sub>, SO<sub>2</sub>, and O<sub>3</sub>.<sup>[34]</sup> In that same study, RH had the greatest impact on the response of the NO<sub>2</sub> sensor in comparison to T, CO, CO<sub>2</sub>, SO<sub>2</sub>, and O<sub>3</sub>.<sup>[34]</sup> In the current study, the linear regression model developed for the Aeroqual NO<sub>2</sub> sensor showed that the effect of O<sub>3</sub> on the performance of the Aeroqual NO<sub>2</sub> sensor was the greatest (i.e., greater magnitude of regression coefficient for O<sub>3</sub> in comparison to RH). This difference between the current

study and that of Zimmerman et al.<sup>[34]</sup> could be due to the use of different NO<sub>2</sub> sensors (i.e., Alphasense ID: NO<sub>2</sub>-B43F versus the Aeroqual NO<sub>2</sub> sensor) that have different levels of sensitivity to cofactors (RH and O<sub>3</sub>, for example). RH values in the study by Zimmerman et al. was not reported, however, O<sub>3</sub> concentration was approximately 0–42 ppb,<sup>[34]</sup> and 1–65 ppb in the current study.

Other researchers have also reported the effects of cofactors of electrochemical sensor performance in the field.<sup>[32,33]</sup> Although controlling the influence of cofactors of sensor performance has been shown to improve the correlation of the corrected sensor measurements to reference/equivalent instruments, previous studies did not include comparison to the NIOSH  $\pm 25\%$  accuracy criterion.<sup>[32–34]</sup> Therefore, it is unclear whether the tested sensors in the quoted studies<sup>[32–34]</sup> are suitable for the use of occupational exposure assessment. Furthermore, the practical benefit of the sensors for exposure assessment may not be economical, given that measurements from other monitoring devices such as reference instruments are required to obtain accurate measurements using corrective models. To obtain accurate measurements from so-called low-cost sensors, modifications in hardware and operating technique may be required.

## Limitations

We used relatively short co-locating periods for the instruments. Nevertheless, hourly fluctuations of T and RH during the three co-locating periods were representative of different weather conditions (for the seasons that we measured). As a result, we were able to assess the accuracy of the sensors in the field during different representative outdoor conditions. Restricting our analysis to measurements  $\geq$  LOD of the tested sensors can potentially cause a selection bias where data analyzed are not representative of environmental conditions associated with very low (i.e.,  $<$  LOD) sensor measurements. However, this method was employed in the current study in order to attain an unequivocal assessment of the performance of the tested sensors in the field. The operators of the monitoring station were only able to provide hourly averages of NO<sub>2</sub> and O<sub>3</sub>, and for this reason the removal of measurements  $<$  LOD were made after calculating hourly averages. The removal of raw measurements (i.e., data logged per minute)  $<$  LOD is a more conservative approach to ensure the validity of data. Finally, the only potentially interfering cofactors that we measured were RH and T; there may be additional contributions from other atmospheric pollutants.



## Conclusions

Raw measurements from the Aeroqual O<sub>3</sub> and NO<sub>2</sub> and Cairclip O<sub>3</sub>/NO<sub>2</sub> sensors may be unsuitable for exposure assessment of outdoor workers such as traffic controllers, toll booth workers, and commercial drivers. Utilizing models to correct for cofactors of sensor performance are required to ensure accurate occupational exposure assessment. The practical benefit of these so-called low-cost sensors may not be economical, given that measurements from other monitoring devices such as reference/equivalent instruments are required to obtain accurate measurements from corrective models.

## Recommendations

We recommend co-locating NO<sub>2</sub> and O<sub>3</sub> sensors with validated reference instruments to investigate the accuracy of sensor measurements before using them for occupational exposure assessment.

## Funding

The sensors used in this study were purchased from grants received from the University of Cincinnati Center for Environmental Genetics (National Institute of Environmental Health Sciences; 5P30 ES006096-22) and the University of Cincinnati Education Research Center (National Institute for Occupational Safety and Health; T42OH008432). K.I. was supported by the Healthy Homes Technical Studies Grant OHHH0027-14 (United States Department of Housing and Urban Development) and the University of Cincinnati, Department of Environmental Health Graduate Assistant Scholarship.

## Acknowledgments

We thank Chris Harrison, Christina Boss and Molnar Eve of the Southwest Ohio Air Quality Agency for assisting us with the deployment of the sensors and the collection of data from the monitoring station.

## References

- [1] Roth, H.D., P.M.T. Hwang, and Y. Li: Assessment of recent ozone short-term epidemiologic studies. *Inhal. Toxicol.* 13(1):1–24 (2001).
- [2] Neidell, M.: Information, avoidance behavior, and health the effect of ozone on asthma hospitalizations. *J. Hum. Resour.* 44(2):450–478 (2009).
- [3] Chen, T.-M., W.G. Kuschner, J. Gokhale, and S. Shofer: Outdoor air pollution: Ozone health effects. *Amer. J. Med. Sci.* 333(4):244–248 (2007).
- [4] Latza, U., S. Gerdes, and X. Baur: Effects of nitrogen dioxide on human health: Systematic review of experimental and epidemiological studies conducted between 2002 and 2006. *Int. J. Hyg. Environ. Health* 212(3):271–287 (2009).
- [5] Samoli, E., E. Aga, G. Touloumi, et al.: Short-term effects of nitrogen dioxide on mortality: An analysis within the aphea project. *Eur. Respir. J.* 27(6):1129–1138 (2006).
- [6] Gillespie-Bennett, J., N. Pierse, K. Wickens, J. Crane, and P. Howden-Chapman: The respiratory health effects of nitrogen dioxide in children with asthma. *Eur. Respir. J.* 38(2):303–309 (2011).
- [7] Ghozikali, M.G., M. Mosafari, G.H. Safari, and J. Jaafari: Effect of exposure to O<sub>2</sub>, NO<sub>2</sub>, SO<sub>2</sub> on chronic obstructive pulmonary disease hospitalizations in Tabriz, Iran. *Environ. Sci. Pollut. Res.* 22(4):2817–2823 (2015).
- [8] Jerrett, M., R.T. Burnett, C.A. Pope III, et al.: Long-term ozone exposure and mortality. *New Eng. J. Med.* 360(11):1085–1095 (2009).
- [9] Carslaw, D.C., and S.D. Beevers: Estimations of road vehicle primary NO<sub>2</sub> exhaust emission fractions using monitoring data in London. *Atmos. Environ.* 39(1):167–177 (2005).
- [10] Finlayson-Pitts, B.J., and J.N. Pitts: Tropospheric air pollution: Ozone, airborne toxics, polycyclic aromatic hydrocarbons, and particles. *Science* 276(5315):1045–1051 (1997).
- [11] Jacob, D.J.: Heterogeneous chemistry and tropospheric ozone. *Atmos. Environ.* 34(12):2131–2159 (2000).
- [12] Wongsurakiat, P., K. Maranetra, A. Nana, C. Naruman, M. Aksornint, and T. Chalermpanyakorn: Respiratory symptoms and pulmonary function of traffic policemen in Thonburi. *J. Med. Assoc. Thailand = Chotmaiher Thangphaet* 82(5):435–443 (1999).
- [13] Alexeeff, S.E., A.A. Litonjua, H. Suh, D. Sparrow, P.S. Vokonas, and J. Schwartz: Ozone exposure and lung function: Effect modified by obesity and airways hyperresponsiveness in the VA normative aging study. *Chest* 132(6):1890–1897 (2007).
- [14] Pal, P., A.J. Robert, T. Dutta, and G. Pal: Pulmonary function test in traffic police personnel in Pondicherry. *Ind. J. Physiol. Pharmacol.* 54(4):329–336 (2010).
- [15] Sehgal, M., R. Suresh, V.P. Sharma, and S.K. Gautam: Assessment of outdoor workers' exposure to air pollution in Delhi (India). *Int. J. Environ. Stud.* 72(1):99–116 (2015).
- [16] Lewne, M., G. Nise, M.-L. Lind, and P. Gustavsson: Exposure to particles and nitrogen dioxide among taxi, bus and lorry drivers. *Int. Arch. Occup. Environ. Health* 79(3):220–226 (2006).
- [17] Chawla, A., and A. Lavania: Air pollution and fuel vapour induced changes in lung functions: Are fuel handlers safe? *Ind. J. Physiol. Pharmacol.* 52(3):255–261 (2008).

- [18] **Bastress, E.K.:** Impact of aircraft exhaust emissions at airports. *Environ. Sci. Technol.* 7(9):811–816 (1973).
- [19] **Lewné, M., N. Plato, and P. Gustavsson:** Exposure to particles, elemental carbon and nitrogen dioxide in workers exposed to motor exhaust. *Ann. Occup. Hyg.* 51(8):693–701 (2007).
- [20] **Raaschou-Nielsen, O., M.L. Nielsen, and J. Gehl:** Traffic-related air pollution: exposure and health effects in Copenhagen street cleaners and cemetery workers. *Arch. Environ. Health Int. J.* 50(3):207–213 (1995).
- [21] **Lam, G., D. Leung, M. Niewiadomski, S. Pang, A. Lee, and P. Louie:** Street-level concentrations of nitrogen dioxide and suspended particulate matter in Hong Kong. *Atmos. Environ.* 33(1):1–11 (1998).
- [22] **American Conference of Governmental Industrial Hygienists:** *Nitrogen Dioxide TLV®. Chemical Substances.* 7th Edition Documentation, ACGIH (2012).
- [23] **American Conference of Governmental Industrial Hygienists:** *Ozone TLV®. Chemical Substances.* 7th Edition Documentation, ACGIH (2001).
- [24] **Brauer, M., and J.D. Spengler:** Nitrogen dioxide exposures inside ice skating rinks. *Am. J. Publ. Health* 84(3):429–433 (1994).
- [25] **Levy, J.I., K. Lee, Y. Yanagisawa, P. Hutchinson, and J.D. Spengler:** Determinants of nitrogen dioxide concentrations in indoor ice skating rinks. *Am. J. Publ. Health* 88(12):1781–1786 (1998).
- [26] **Olin, A.-C., G. Granung, S. Hagberg, et al.:** Respiratory health among bleachery workers exposed to ozone and chlorine dioxide. *Scand. J. Work Environ. Health* 117–123 (2002).
- [27] **National Institute For Occupational Safety And Health:** “Nitrogen Dioxide (Diffusive Sampler).” NIOSH (1998). Available at <https://www.cdc.gov/niosh/docs/2003-154/pdfs/6700.pdf> (accessed July 31, 2018).
- [28] **Occupational Safety and Health Administration:** *Ozone In Workplace Atmospheres (Impregnated Glass Fiber Filter)*, (Report #OSHA Id-214), Branch of Inorganic Methods Development, March 1995. Available at <https://www.osha.gov/dts/sltc/methods/inorganic/id214/id214.pdf> (accessed July 31, 2018).
- [29] **Lewis, A.C., J.D. Lee, P.M. Edwards, et al.:** Evaluating the performance of low cost chemical sensors for air pollution research. *Faraday Discuss.* 189:85–103 (2016).
- [30] **Lewis, A., and P. Edwards:** Validate personal air-pollution sensors: Alastair Lewis and Peter Edwards call on researchers to test the accuracy of low-cost monitoring devices before regulators are flooded with questionable air-quality data. *Nature* 535(7610):29–32 (2016).
- [31] **Duvall, R.M., R.W. Long, M.R. Beaver, K.G. Kronmiller, M.L. Wheeler, and J.J. Szykman:** Performance evaluation and community application of low-cost sensors for ozone and nitrogen dioxide. *Sensors* 16(10):1698 (2016).
- [32] **Cross, E.S., L.R. Williams, D.K. Lewis, et al.:** Use of electrochemical sensors for measurement of air pollution: Correcting interference response and validating measurements. *Atmos. Measure. Techniq.* 10(9):3575 (2017).
- [33] **Jiao, W., G. Hagler, R. Williams, et al.:** Community Air Sensor Network (CAIRSENSE) Project: Evaluation of low-cost sensor performance in a suburban environment in the Southeastern United States. *Atmos. Measure. Techniq.* 9(11):5281 (2016).
- [34] **Zimmerman, N., A.A. Presto, S.P. Kumar, et al.:** A machine learning calibration model using random forests to improve sensor performance for lower-cost air quality monitoring. *Atmos. Measure. Techniq.* 11(1):291 (2018).
- [35] **Tarald, O.K.:** Cautionary note about  $R^2$ . *Amer. Statistician* 39(4 Pt 1):279–285 (1985).
- [36] **National Institute For Occupational Safety and Health:** *Components for Evaluation of Direct-Reading Monitors for Gases and Vapors*, (DHHS Publication No. 2012-162). Department of Health and Human Services, Centers for Disease Control and Prevention, National Institute for Occupational Safety and Health, July 2012. Available at <https://www.cdc.gov/niosh/docs/2012-162/pdfs/2012-162.pdf> (accessed July 31, 2018).
- [37] **U.S. Environmental Protection Agency:** “National Ambient Air Quality Standards (NAAQS).” Available at <https://www.epa.gov/sites/production/files/2015-02/documents/criteria.pdf> (accessed July 31, 2018).
- [38] **European Commission:** “Air Quality Standards.” Available at <http://ec.europa.eu/environment/air/quality/standards.htm> (accessed July 31, 2018).
- [39] **U.S. Environmental Protection Agency:** “Quality Assurance Requirements for Monitors Used in Evaluations of National Ambient Air Quality Standards,” 40 Cfr Appendix A4 (July 1, 2016).
- [40] **U.S. Environmental Protection Agency:** “List of Designated Reference and Equivalent Methods: National Exposure Research Laboratory, Office Of Research And Development,” 2017. Available at [https://www3.epa.gov/ttnamtl1/files/ambient/criteria/AMTIC\\_List\\_June\\_2017\\_update\\_6-19-2017.pdf](https://www3.epa.gov/ttnamtl1/files/ambient/criteria/AMTIC_List_June_2017_update_6-19-2017.pdf) (accessed July 31, 2018).
- [41] **U.S. Environmental Protection Agency:** *Performance of the Proposed New Federal Reference Method for Measuring Ozone Concentrations in Ambient Air*. (Report #EPA/600/R-14/432 (NTIS PB2015-101240) Office of Research and Development, National Exposure Research Laboratory, November 2014. Available at [https://cfpub.epa.gov/si/si\\_public\\_record\\_report.cfm?direntryid=293939](https://cfpub.epa.gov/si/si_public_record_report.cfm?direntryid=293939) (accessed July 31, 2018).
- [42] **U.S. Environmental Protection Agency:** *Cairpol Cairclip O<sub>3</sub>-NO<sub>2</sub>*, by R. Williams, A. Kaufman, S. Garvey (Report #EPA/600/R-15/117). Office of Research and Development, National Exposure Research Laboratory, May 2015. Available at <https://nepis.epa.gov/Exe/ZyPDF.cgi/P100MOXD.PDF?Dockey=P100MOXD.PDF> (accessed July 31, 2018).
- [43] **Aeroqual:** “Nitrogen Dioxide Sensor 0-1ppm.” Available at <https://www.aeroqual.com/product/nitrogen-dioxide-sensor-0-1ppm> (accessed, July 31, 2018).

- [44] **Aeroqual**: "Ozone Sensor 0-0.15ppm." Available at <https://www.aeroqual.com/product/ozone-sensor-ozu> (accessed July 31, 2018).
- [45] **MSA Gas Monitors**: "MSA Altair Pro Single-Gas Detector - Nitrogen Dioxide (NO<sub>2</sub>) - 10076731." Available at <https://www.msagasmonitors.com/10076731.html> (accessed, July 31, 2018).
- [46] **Dräger**: "Electrochemical Drägersensor Sensors for Fixed Gas Detectors." Available at <https://www.draeger.com/products/content/sensor-ec-pi-9045596-en-master.pdf> (accessed July 31, 2018).
- [47] **Bard, A.J. and L.F. Faulkner**: *Electrochemical Methods: Fundamental and Applications*. New York: John Wiley & Sons, Inc., 2001.
- [48] **Chang, S.C., and J.R. Stetter**: Electrochemical NO<sub>2</sub> gas sensors: Model and mechanism for the electroreduction of NO<sub>2</sub>. *Electroanalysis* 2(5):359–365 (1990).
- [49] **Fine, G.F., L.M. Cavanagh, A. Afonja, and R. Binions**: Metal oxide semi-conductor gas sensors in environmental monitoring. *Sensors* 10(6):5469–5502 (2010).
- [50] **Seiyama, T., A. Kato, K. Fujiishi, and M. Nagatani**: A new detector for gaseous components using semi-conductive thin films. *Analyt. Chem.* 34(11):1502–1503 (1962).
- [51] **Peterson, P.J., A. Aujla, K.H. Grant, et al.**: Practical use of metal oxide semiconductor gas sensors for measuring nitrogen dioxide and ozone in urban environments. *Sensors* 17(7):1653 (2017).
- [52] **Cairpol**: "Technical Datasheet CAIRSENS O3/NO<sub>2</sub>." Available at <http://cairpol.com/wp-content/uploads/2017/05/cairsens-O3no2-0-250ppb-technical-specifications.pdf> (accessed, July 31, 2018).
- [53] **Teledyne API**: "Operation Manual Model T200 NO/NO<sub>2</sub>/NO<sub>x</sub> Analyzer." San Diego, CA: Teledyne API (TAPI). Available at [http://www.teledyne-api.com/manuals/06858\\_t200.pdf](http://www.teledyne-api.com/manuals/06858_t200.pdf) (accessed July 31, 2018).
- [54] **Teledyne API**: "Operation Manual Model 400E Photometric Ozone Analyzer." San Diego, CA: Teledyne API (TAPI). Available at [http://eservices.teledyne-api.com/manuals/04316f\\_400e.pdf](http://eservices.teledyne-api.com/manuals/04316f_400e.pdf) (accessed July 31, 2018).
- [55] **R Studio**: "Integrated Development for R." Boston, MA: R Studio, Inc., 2016. Available at <http://www.rstudio.com/>.
- [56] **Diciccio, T.J., and B. Efron**: Bootstrap confidence intervals. *Statist. Sci.* 189–212 (1996).
- [57] **Efron, B.**: Better bootstrap confidence intervals. *J. Amer. Statist. Assoc.* 82(397):171–185 (1987).
- [58] **Efron, B., and R. Tibshirani**: Bootstrap methods for standard errors, confidence intervals, and other measures of statistical accuracy. *Statist. Sci.* 54–75 (1986).
- [59] **Wang, F.**: Confidence interval for the mean of non-normal data. *Qual. Reliab. Eng. Int.* 17(4):257–267 (2001).
- [60] **Mead, M.I., O. Popoola, G. Stewart, et al.**: The use of electrochemical sensors for monitoring urban air quality in low-cost, high-density networks. *Atmos. Environ.* 70:186–203 (2013).